

Adsorption-based Process for Separation of Aleuritic Acid from Lac

by

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Dedicated
to
my Parents,
my Ph.D. Supervisors,
&
Prof. A. K. Gupta

CERTIFICATE

This is to certify that the thesis entitled '**Adsorption-based Process for Separation of Aleuritic Acid from Lac**' being submitted by **Ms. Navdeep Kaur** to the Indian Institute of Technology Delhi for award of Doctor of Philosophy is a record of bonafide research work carried out by her under our guidance and supervision in conformity with the rules and regulations of Indian Institute of Technology Delhi.

The research report and results presented in this thesis have not been submitted, in part or full, to any other university or institute for the award of any degree or diploma.

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ABSTRACT

Aleuritic acid (9, 10, 16-Trihydroxy palmitic acid) is a hydroxy fatty acid usually isolated from lac. It is an important starting material for a large number of specialty chemicals required for flavours and fragrances. This thesis investigates the feasibility of an adsorption based process for isolation and recovery of aleuritic acid, because the current processes based on dissolution and re-crystallization result in low yields. In the present work, hitherto unknown adsorption characteristics of aleuritic acid over an anion exchange resin Tulsion A-27 MP have been investigated. The batch equilibrium and kinetic behavior of adsorption of sodium salt of aleuritic acid by the resin has been experimentally studied. Equilibrium adsorption data obtained has been correlated using standard adsorption isotherms. The isotherms show a characteristic dependence on initial concentration of solution. The observations during this study have been rationalized through a model developed on the basis of Langmuir adsorption of the ion on the resin while also accounting for the solution chemistry, hydrolysis and dissociation of different species involved.

Kinetic studies have been performed by monitoring the conductivity of a well-stirred batch of the resin in the adsorbate solution and monitoring the conductivity as a function of time. External mass transport limitations have been eliminated by studying the effect of stirrer RPM. Adsorption kinetics have been investigated for various resin loadings and the data was attempted to be fitted with several simple and more complicated models accounting for the microstructure of resin. These studies are successful in predicting the overall trend but were seen to be limiting in modeling the characteristic times in the process, captured by the slopes and curvatures of the curves. Finally, a modified model for mass transfer to a sphere from a finite solution appeared to work well in this situation and rationalization of the data has been achieved.

Finally column studies for adsorption and desorption, which highlight the feasibility or the use of adsorption on the column have been analyzed and alternate process based on the liquid-solid circulating fluidized bed technology has been proposed for the process. Various design and engineering considerations for the process are discussed.

TABLE OF CONTENTS

CERTIFICATE	i
ACKNOWLEDGEMENTS	ii
ABSTRACT	v
TABLE OF CONTENTS	vii
LIST OF FIGURES	x
LIST OF TABLES	xiii
1. Introduction, background and research objectives	1
1.1 Aleuritic acid and its applications	1
1.2 Present methods of Extraction of Aleuritic Acid	3
1.3 Preliminary experimental work	6
1.4 Motivation	13
1.5 Research objectives	14
1.6 Structure of the thesis	15
2. Batch equilibrium studies	17
2.1 Introduction	17
2.2 Experimental procedure for equilibrium studies	19
2.2.1 Conditioning of resin	19
2.2.2 Solution concentration using titration	20
2.2.3 Solution concentration using conductivity measurements	22
2.2.4 Adsorption equilibrium experiments	24
2.3 Results and discussion	24
2.3.1 Adsorption equilibria	24

2.3.2	Rigorous analysis and modeling	35
2.4	Summary and conclusions	45
3.	Batch kinetic studies	46
3.1	Background	46
3.2	Intraparticle in ion-exchange resins	49
3.3	Experimental procedure for kinetic studies	51
3.4	Results and discussion	53
3.4.1	Effect of stirrer speed: External mass transfer resistance	53
3.4.2	General trends in kinetic data	54
3.4.3	Homogeneous diffusion-adsorption model: Rectangular boundary condition	60
3.4.4	Homogeneous diffusion-adsorption model: Langmuir boundary condition	63
3.4.5	Macropore-Microsphere diffusion adsorption model	66
3.4.6	Homogeneous sphere in finite solution control	70
3.5	Summary and conclusions	79
4.	Batch desorption studies and column studies	81
4.1	Desorption and recovery of aleuritic acid in batch systems	81
4.1.1	Experimental methodology	82
4.1.2	Results and discussion	84
4.2	Column studies: Adsorption and desorption of pure aleuritic acid	89
4.2.1	Methodology used in experiments	90
4.2.2	Results and discussion	93

4.3	Summary and conclusions	97
5.	Process Conceptualization	99
5.1	Considerations for process conceptualization	99
5.2	Proposed contacting pattern for adsorber-desorber	102
5.3	Hydrodynamic feasibility	105
5.4	Proposed process	108
5.5	Preliminary Design Calculations	109
6	Summary and conclusions	124
6.1	Summary and conclusions	124
6.2	Scope for future work	126
	Appendix A	127
	Notations	143
	References	148