

1D Electrospun Nanofibers Supported Metal Oxide-based Nanostructures for Supercapacitor Applications

Vaishali Tanwar



**Department of Chemistry
Indian Institute of Technology Delhi**

October 2024

© Indian Institute of Technology Delhi (IITD), New Delhi, 2024

1D Electrospun Nanofibers Supported Metal Oxide-based Nanostructures for Supercapacitor Applications

**by
Vaishali Tanwar
Department of Chemistry**

Submitted

in fulfilment of the requirements of the degree of

Doctor of Philosophy

to the



Indian Institute of Technology Delhi

October 2024

*Dedicated to my
Family*

CERTIFICATE

This is to certify that the thesis entitled “**1D Electrospun Nanofibers Supported Metal Oxide-based Nanostructures for Supercapacitor Applications,**” being submitted by **Ms. Vaishali Tanwar** to the **Indian Institute of Technology, Delhi**, for the award of the degree of **Doctor of Philosophy**, is a record of bonafide research work carried out by her. She has worked under my supervision and has fulfilled the requirements, which, to my knowledge, has reached the requisite standard for submitting the thesis. The results in this thesis have not been submitted to any other university or institute, whether in part or full, for the awards of any other degree or diploma.

Prof. Pravin P. Ingole

Professor

Department of Chemistry

Indian Institute of Technology

ACKNOWLEDGMENTS

Embarking on the Ph.D. journey has been profoundly transformative, and I owe immense gratitude to numerous individuals who supported me along the way.

First and foremost, I extend my deepest thanks to my supervisor, **Prof. Pravin P Ingole**. Your valuable insights, constant patience, and sustained support have been the bedrock of this thesis. Your ability to inspire and challenge me has shaped my academic growth and fostered my personal development in ways I cannot overstate. Your thorough review of my work and constructive feedback during challenging times have been invaluable.

I would also like to extend my gratitude to my doctoral committee members, Prof. Ashok K Ganguli, Prof. Sameer Sapra, and Prof. Bhanu Nandan. Your collective expertise, insightful comments, and suggestions have significantly enriched this research. I immensely thank the DST-Inspire and MHRD for their generous financial support. Your contributions have provided the essential resources and opportunities for this research. I also thank the Central Research Facility (CRF), NanoScale Research Facility (NRF), and Departmental Instrumentation Facility at IIT Delhi for providing research facilities. I would like to acknowledge the heads of the Chemistry department during my tenure at IIT Delhi: Prof. Ashok K. Ganguli, Prof. Anil J. Elias, Prof. Narayanan Kurur, Prof. Sidharth Pandey, and Prof. S Nagendran.

I am deeply grateful to all my collaborators at IITD. Special thanks to Dr. Leena Nebhnai (Department of Materials Science and Engineering, IITD) and Dr. Rasmita Barik, former post-doc in the lab, for their generous and continuous support throughout this journey. I would also like to thank all current and former labmates: **Priya**, Jyoti, Paras, Cini, **Ajay**, **Aamir**, Aejaz, Sagar, Navneet, Shivangi, Aditi, Jagriti and Bhawna, **Saurabh Kumar Pathak**, Aditi Ashok Gujare, **Rajat**, Sandeep, Baldeep, Ankita, Sonu, Aastha, Aman and Satyam for providing a comfortable, helpful, inventive atmosphere that made my lab days enjoyable and fun. I further thank some former postdocs and senior members in the lab, Dr. Shahla, Dr. Kashyap, Dr. Priyanka, Dr. Harsha, Dr. Nusrat, Dr. Soniya, Dr. Shwetambara Jha, Dr. Preeti for generously sharing their research experiences with me.

A special mention goes to my family for their unconditional love and encouragement. To my mother, **Lalita Tanwar**, I thank her for instilling in me the values of hard work and perseverance. Her continuous belief in my abilities and love has been my guiding force. To my brothers, **Harsh Tanwar** and **Anup Tanwar**, their constant support and encouragement have

been invaluable. To my husband, **Ankit**, his unwavering love, patience, and support have been my anchor throughout this journey. I thank him for standing by my side, believing in me, providing the emotional strength I needed to persevere, and hearing my sob stories every time I hit rock bottom. His understanding and encouragement have been instrumental in my success. Lastly, I would like to acknowledge my best friends, Shilpa and Vaishali, who are outside the academic world. The friendship and the moments of joy and laughter we shared have provided me with the balance needed to navigate the challenges of this journey.

Thank you all for your invaluable contributions to the successful completion of this thesis. Your support has been instrumental in my development as an individual and a researcher, and I am forever grateful.

Vaishali Tanwar

ABSTRACT

The rapid depletion of conventional energy resources, coupled with escalating environmental concerns and increasing global energy demand, has elevated the pursuit of clean and renewable energy alternatives. However, the intermittent availability of renewable energy sources and the growing consumer market highlight the crucial need to develop sustainably advanced energy storage systems capable of efficiently meeting peak power demands. Recently, Electrochemical supercapacitors (SCs) have gained considerable attention due to their superior energy densities compared to conventional capacitors, as well as higher power densities, longer cycle life, and enhanced safety features compared to batteries. They are emerging as promising solutions for substantial and advanced storage systems, with potential applications in electric vehicles, backup power sources, and pulse power providers, poised to attract significant consumer interest in the coming years. Despite these advancements, challenges such as inferior energy density ($<10 \text{ Wh kg}^{-1}$), limited nanostructured electrode designs and scalability issues hinder their widespread commercialization. To address these challenges, this thesis aims to synthesize innovative nanoarchitectures, particularly 1D nanofibers supported metal oxide (MO) based composites, for enhanced electrochemical performance. This includes achieving high energy densities while maintaining robust power densities, thereby advancing the practicality and application of SCs in various energy storage applications. The thesis systemically focuses on the nanoscale engineering of electrode materials using scalable approaches such as electrospinning to generate advanced hybrid materials with optimized morphology and composition. It is structured into eight chapters.

Chapter 1 provides a comprehensive overview of global energy scenarios and the various energy storage systems that have been widely studied, highlighting their respective advantages and disadvantages. It introduces supercapacitors (SCs), explains their fundamental operational principles and components, and discusses their classification and energy storage mechanisms. The chapter extensively reviews the literature on carbon-based materials, metal oxides, and their hybrid composites as electrodes for SCs, focusing on their electrochemical performance. Special attention is given to developing electrode materials using scalable approaches such as electrospinning and subsequent treatments to create highly efficient hybrid structures of metal oxides and polymer fibers. The chapter also outlines the motivation for the thesis work and articulates its objectives clearly.

Chapter 2 details the methodology for synthesizing 1D nanofiber-supported metal oxide-based composites. Additionally, the chapter outlines the experimental methodologies employed for

material investigation and highlights various characterization techniques crucial for establishing structure-property relationships. It provides a comprehensive explanation of the electrochemical methodologies used to evaluate the electrochemical performance, discussing key parameters for assessing the supercapacitive performance of the synthesized materials.

Chapter 3 introduces W-PAN, a high-performance hybrid electrode material fabricated by decorating 1D electrospun PAN nanofibers with WO_3 nano petals. This approach enhances WO_3 's electrochemical capabilities through a tailored support structure. The study elucidates the charge storage mechanisms of W-PAN, highlighting synergies between WO_3 and PAN fibers that lead to superior electrochemical properties. W-PAN exhibits extended discharge times, higher specific capacitance (C_{sp}), and enhanced cyclic durability compared to pristine WO_3 and PAN. The fibrous PAN matrix supports WO_3 deposition, facilitating strong interactions and contributing to its pseudocapacitive behavior. Comprehensive analysis reveals dual charge storage mechanisms-double-layer and pseudocapacitive underscoring W-PAN's effectiveness in energy storage.

Chapter 4 meticulously details the synthesis and optimization of WO_3 nanoparticles grown onto PVP-modified carbon nanofibers (m-CNFs) to create advanced hybrid supercapacitor materials. The chapter discusses the strategic use of phase-separation mediated electrospinning of PAN and PVP polymers, followed by controlled thermal treatment, to prepare m-CNFs. This approach allows for precise PAN to PVP ratio adjustment, providing insights into how the polymer-derived supports influence ion transport efficiency and enhance the utilization of electrode materials. Key findings highlight the significant role of m-CNFs in establishing intimate interfaces with WO_3 nanoparticles. This structural integration enhances electron and ion transfer rates, thereby boosting the pseudocapacitive charge storage capacity of the hybrid material.

Chapter 5 explores the development of hierarchical multi-metal oxide (WO_3/SnO_2) @carbon-based nanostructures as advanced supercapacitor electrode materials. The chapter emphasizes the strategic design of these nanostructures to enhance charge storage capabilities through induced lattice defects and heterointerface interactions. The synthesis method involves single-spinneret electrospinning followed by controlled calcination, ensuring a scalable and efficient approach to fabricate WO_3/SnO_2 @C hybrids. It also highlights the superior electrochemical performance of WO_3/SnO_2 @C compared to individual WO_3 and SnO_2 materials. This exceptional electrochemical performance is attributed to the synergistic effects of the hierarchical morphology and induced lattice defects, which facilitate efficient charge transfer and minimize degradation over extended cycling. The chapter also elucidates the charge

storage mechanisms of $\text{WO}_3/\text{SnO}_2@\text{C}$, emphasizing the capacitive adsorption/desorption of H^+ ions as the primary contributor to its electrochemical capacitance.

Chapter 6 comprehensively explores the synthesis strategy, structural characteristics, and electrochemical performance of 1D SnO_2 - fibers. The chapter highlights the synthesis process involving electrospinning a SnO_2 precursor solution with polyvinylpyrrolidone (PVP). This approach enables the formation of uniform and highly porous 1D nanostructures of SnO_2 , which is crucial for optimizing the material's electrochemical performance. Key to the methodology is PVP as a soft carbon template in facilitating the creation of a fibrous morphology with a high surface area composed of densely packed small spheroidal nanoparticles of SnO_2 . This structural configuration enhances the active electrochemical surface area, promoting efficient ion intercalation and de-intercalation processes at the electrode/electrolyte interface during charge-discharge cycles. The study demonstrates that SnO_2 nanocomposites are promising electrode materials, offering ultra-high energy and power densities suitable for various practical energy storage needs.

Chapter 7 explores a novel approach to enhance the energy storage capabilities of hybrid electrode materials by synthesizing bimetallic oxides (Bi_2O_3 and SnO_2) embedded in carbon fibers (CFs), termed BS/NCFs-based 1D nanostructures. The chapter details a straightforward method involving the electrospinning of polyacrylonitrile (PAN) combined with a metal precursor mixture, followed by controlled calcination to achieve the desired composite structure. A strategic PAN carbonization facilitates the in-situ nucleation of Bi_2O_3 and SnO_2 nanoparticles within the CF matrix. A highly porous 1D structure characterized by mesoporous pore sizes and a high specific surface area enhances electrolyte accessibility, promotes rapid electron transport, and facilitates effective charge storage mechanisms crucial for supercapacitive performance. The chapter emphasizes the structure-property relationships of BS/NCFs, highlighting the benefits of optimal MO composition and calcination temperature in achieving enhanced electrochemical properties.

Chapter 8 serves as the culmination of the thesis, presenting a comprehensive summary and conclusion of the investigations conducted throughout the study. This final chapter is pivotal in synthesizing the key findings, discussing their implications, and outlining future prospects for advancing efficient SC electrode materials.

अतिक्रमण

पारंपरिक ऊर्जा संसाधनों की तेजी से कमी, बढ़ती पर्यावरणीय चिंताओं और बढ़ती वैश्विक ऊर्जा मांग ने स्वच्छ और नवीकरणीय ऊर्जा विकल्पों की खोज को बढ़ा दिया है। हालांकि, अक्षय ऊर्जा स्रोतों की रुक-रुक कर उपलब्धता और बढ़ता उपभोक्ता बाजार बिजली की चरम मांगों को कुशलता से पूरा करने में सक्षम स्थायी रूप से उन्नत ऊर्जा भंडारण प्रणालियों को विकसित करने की महत्वपूर्ण आवश्यकता को उजागर करता है। हाल ही में, इलेक्ट्रोकेमिकल सुपरकैपेसिटर (SC) ने पारंपरिक कैपेसिटर की तुलना में अपनी बेहतर ऊर्जा घनत्व के साथ-साथ उच्च शक्ति घनत्व, लंबे चक्र जीवन और बैटरी की तुलना में बेहतर सुरक्षा सुविधाओं के कारण काफी ध्यान आकर्षित किया है। वे इलेक्ट्रिक वाहनों, बैकअप बिजली स्रोतों और पल्स बिजली प्रदाताओं में संभावित अनुप्रयोगों के साथ पर्याप्त और उन्नत भंडारण प्रणालियों के लिए आशाजनक समाधान के रूप में उभर रहे हैं, जो आने वाले वर्षों में महत्वपूर्ण उपभोक्ता रुचि को आकर्षित करने के लिए तैयार हैं। इन प्रगति के बावजूद, निम्न ऊर्जा घनत्व ($<10 \text{ Wh kg}^{-1}$) सीमित नैनोस्ट्रक्चर्ड इलेक्ट्रोड डिजाइन और मापनीयता के मुद्दे जैसी चुनौतियां उनके व्यापक व्यावसायीकरण में बाधा डालती हैं। इन चुनौतियों का समाधान करने के लिए, इस शोध प्रबंध का उद्देश्य विद्युत रासायनिक प्रदर्शन को बढ़ाने के लिए नवीन नैनोआर्किटेक्चर, विशेष रूप से 1डी नैनोफाइबर समर्थित धातु ऑक्साइड (MO) आधारित कंपोजिट को संश्लेषित करना है। इसमें मजबूत बिजली घनत्व बनाए रखते हुए उच्च ऊर्जा घनत्व प्राप्त करना शामिल है, जिससे विभिन्न ऊर्जा भंडारण अनुप्रयोगों में अनुसूचित जातियों की व्यावहारिकता और अनुप्रयोग को आगे बढ़ाया जा सकता है। थीसिस व्यवस्थित रूप से इलेक्ट्रोड सामग्री के नैनोस्केल इंजीनियरिंग पर केंद्रित है, जिसमें इष्टतम आकृति विज्ञान और संरचना के साथ उन्नत संकर सामग्री उत्पन्न करने के लिए इलेक्ट्रोस्पिनिंग जैसे स्केलेबल दृष्टिकोण का उपयोग किया जाता है। इसे आठ अध्यायों में संरचित किया गया है।

अध्याय 1 वैश्विक ऊर्जा परिदृश्यों और विभिन्न ऊर्जा भंडारण प्रणालियों का एक व्यापक अवलोकन प्रदान करता है जिनका व्यापक रूप से अध्ययन किया गया है, उनके संबंधित लाभ और नुकसान को उजागर करता है। यह सुपरकैपेसिटर (SC) का परिचय देता है जो उनके मौलिक परिचालन सिद्धांतों और घटकों की व्याख्या करता है, और उनके वर्गीकरण और ऊर्जा भंडारण तंत्र पर चर्चा करता है। इस अध्याय में कार्बन-आधारित सामग्रियों, धातु ऑक्साइड और उनके संकर यौगिकों पर साहित्य की व्यापक रूप से समीक्षा की गई है, जो अनुसूचित जातियों के लिए इलेक्ट्रोड के रूप में उनके विद्युत रासायनिक प्रदर्शन पर ध्यान केंद्रित करते हैं। धातु ऑक्साइड और पॉलिमर फाइबर की अत्यधिक कुशल संकर संरचनाओं को बनाने के लिए इलेक्ट्रोस्पिनिंग और बाद के उपचारों जैसे स्केलेबल दृष्टिकोण का उपयोग करके इलेक्ट्रोड सामग्री विकसित करने पर विशेष ध्यान दिया जाता है। यह अध्याय शोध प्रबंध कार्य के लिए प्रेरणा को भी रेखांकित करता है और इसके उद्देश्यों को स्पष्ट रूप से व्यक्त करता है।

अध्याय 2 में 1D नैनोफाइबर-समर्थित धातु ऑक्साइड-आधारित यौगिकों के संश्लेषण की कार्यप्रणाली का विवरण दिया गया है। इसके अतिरिक्त, अध्याय सामग्री जांच के लिए नियोजित प्रयोगात्मक पद्धतियों की रूपरेखा तैयार करता है और संरचना-संपत्ति संबंधों को स्थापित करने के लिए महत्वपूर्ण विभिन्न लक्षण वर्णन तकनीकों पर प्रकाश डालता है। यह इलेक्ट्रोकेमिकल प्रदर्शन का मूल्यांकन करने के लिए उपयोग की जाने वाली इलेक्ट्रोकेमिकल पद्धतियों की एक व्यापक व्याख्या प्रदान करता है, जिसमें संश्लेषित सामग्री के सुपरकैपेसिटिव प्रदर्शन का आकलन करने के लिए प्रमुख मापदंडों पर चर्चा की जाती है।

अध्याय 3 में W-PAN का परिचय दिया गया है, जो एक उच्च प्रदर्शन वाली संकर इलेक्ट्रोड सामग्री है जिसे WO_3 नैनो पंखुड़ियों के साथ 1डी इलेक्ट्रोस्पिन पैन नैनोफाइबर को सजाकर बनाया गया है। यह दृष्टिकोण एक अनुरूप समर्थन संरचना के माध्यम से WO_3 की विद्युत रासायनिक क्षमताओं को बढ़ाता है। अध्ययन डब्ल्यू-पैन के चार्ज भंडारण तंत्र को स्पष्ट करता है, जो WO_3 और PAN फाइबर के बीच तालमेल को उजागर करता है जो बेहतर विद्युत रासायनिक गुणों की ओर ले जाता है। W-PAN प्राचीन WO_3 और W-PAN की तुलना में विस्तारित निर्वहन समय, उच्च विशिष्ट धारिता (C_{sp}) और बढ़ी हुई चक्रीय स्थायित्व प्रदर्शित करता है। रेशेदार PAN मैट्रिक्स WO_3 निक्षेपण का समर्थन करता है, जो मजबूत अंतःक्रिया को सुविधाजनक बनाता है और इसके छद्म कैपेसिटिव व्यवहार में योगदान देता है। व्यापक विश्लेषण से दोहरे चार्ज भंडारण तंत्र-दोहरी-परत और छद्म कैपेसिटिव का पता चलता है जो ऊर्जा भंडारण में डब्ल्यू-पैन की प्रभावशीलता को रेखांकित करता है।

अध्याय 4 उन्नत संकर सुपरकैपेसिटर सामग्री बनाने के लिए पीवीपी-संशोधित कार्बन नैनोफाइबर (m-CNFs) पर उगाए गए WO_3 नैनोकणों के संश्लेषण और अनुकूलन का सावधानीपूर्वक विवरण देता है। इस अध्याय में m-CNFs तैयार करने के लिए PAN और PVP पॉलिमर के चरण-पृथक्करण मध्यस्थता इलेक्ट्रोस्पिनिंग के रणनीतिक उपयोग पर चर्चा की गई है, जिसके बाद नियंत्रित थर्मल उपचार किया गया है। यह दृष्टिकोण सटीक पैन से पीवीपी अनुपात समायोजन की अनुमति देता है, जो इस बात की अंतर्दृष्टि प्रदान करता है कि बहुलक-व्युत्पन्न समर्थन आयन परिवहन दक्षता को कैसे प्रभावित करता है और इलेक्ट्रोड सामग्री के उपयोग को बढ़ाता है। प्रमुख निष्कर्ष WO_3 नैनोपार्टिकल्स के साथ अंतरंग इंटरफेस स्थापित करने में m-CNFs की महत्वपूर्ण भूमिका पर प्रकाश डालते हैं। यह संरचनात्मक एकीकरण इलेक्ट्रॉन और आयन स्थानांतरण दरों को बढ़ाता है, जिससे संकर सामग्री की छद्म कैपेसिटिव चार्ज भंडारण क्षमता को बढ़ावा मिलता है।

अध्याय 5 उन्नत सुपरकैपेसिटर इलेक्ट्रोड सामग्री के रूप में पदानुक्रमित बहु-धातु ऑक्साइड (WO_3/SnO_2)@C-आधारित नैनोस्ट्रक्चर के विकास की पड़ताल करता है। यह अध्याय प्रेरित जाली दोषों और हेटेरोइंटरफेस इंटरैक्शन के माध्यम से चार्ज स्टोरेज क्षमताओं को बढ़ाने के लिए इन नैनोस्ट्रक्चरों के रणनीतिक डिजाइन पर जोर देता है। संश्लेषण विधि में एकल-स्पिनरेट इलेक्ट्रोस्पिनिंग के बाद नियंत्रित कैल्सीनेशन शामिल है, जो WO_3/SnO_2 @C हाइब्रिड बनाने के लिए एक स्केलेबल और कुशल दृष्टिकोण सुनिश्चित करता है। यह व्यक्तिगत WO_3 और SnO_2 सामग्रियों की तुलना में WO_3/SnO_2 @C के बेहतर विद्युत रासायनिक प्रदर्शन पर भी प्रकाश डालता है। इस असाधारण विद्युत रासायनिक प्रदर्शन का श्रेय पदानुक्रमित आकृति विज्ञान और प्रेरित जाली दोषों के सहक्रियात्मक प्रभावों को दिया जाता है, जो कुशल चार्ज हस्तांतरण की सुविधा प्रदान करते हैं और विस्तारित साइक्लिंग पर क्षरण को कम करते हैं। अध्याय WO_3/SnO_2 @C के चार्ज स्टोरेज तंत्र को भी स्पष्ट करता है, जो इसके विद्युत रासायनिक धारिता में प्राथमिक योगदानकर्ता के रूप में H^+ आयनों के कैपेसिटिव एडसोर्प्शन/डिसोर्प्शन पर जोर देता है।

अध्याय 6 1D एसएनओ2-फाइबर की संश्लेषण रणनीति, संरचनात्मक विशेषताओं और विद्युत रासायनिक प्रदर्शन की व्यापक रूप से पड़ताल करता है। यह अध्याय पॉलीविनाइलपाइरोलिडोन(PVP) के साथ एक SnO_2 अग्रदूत समाधान को इलेक्ट्रोस्पिनिंग से जुड़ी संश्लेषण प्रक्रिया पर प्रकाश डालता है। (PVP). यह दृष्टिकोण SnO_2 के समान और अत्यधिक छिद्रपूर्ण 1D नैनोस्ट्रक्चर के गठन को सक्षम बनाता है, जो सामग्री के विद्युत रासायनिक प्रदर्शन को अनुकूलित करने के लिए महत्वपूर्ण है। SnO_2 के घने पैक किए गए छोटे गोलाकार नैनोकणों से बने उच्च सतह क्षेत्र के साथ एक रेशेदार आकृति विज्ञान के

निर्माण को सुविधाजनक बनाने में एक नरम कार्बन टेम्पलेट के रूप में कार्यप्रणाली की कुंजी पी. वी. पी. है। यह संरचनात्मक विन्यास सक्रिय विद्युत रासायनिक सतह क्षेत्र को बढ़ाता है, चार्ज-निर्वहन चक्रों के दौरान इलेक्ट्रोड/इलेक्ट्रोलाइट इंटरफेस पर कुशल आयन इंटरकेलेशन और डी-इंटरकेलेशन प्रक्रियाओं को बढ़ावा देता है। अध्ययन से पता चलता है कि SnO_2 नैनोकंपोजिट आशाजनक इलेक्ट्रोड सामग्री हैं, जो विभिन्न व्यावहारिक ऊर्जा भंडारण आवश्यकताओं के लिए उपयुक्त अति-उच्च ऊर्जा और शक्ति घनत्व प्रदान करते हैं।

अध्याय 7 कार्बन फाइबर (CFs) में एम्बेडेड बाइमेटेलिक ऑक्साइड (Bi_2O_3 और SnO_2) को संश्लेषित करके संकर इलेक्ट्रोड सामग्री की ऊर्जा भंडारण क्षमताओं को बढ़ाने के लिए एक नए दृष्टिकोण की पड़ताल करता है जिसे बीएस/एनसीएफ-आधारित 1D नैनोस्ट्रक्चर कहा जाता है। अध्याय में एक सीधी विधि का विवरण दिया गया है जिसमें एक धातु अग्रदूत मिश्रण के साथ संयुक्त पॉलीएक्रिलोनाइट्राइल (PAN) की इलेक्ट्रोस्पिनिंग शामिल है, जिसके बाद वांछित समग्र संरचना को प्राप्त करने के लिए नियंत्रित कैल्सीनेशन किया जाता है। एक रणनीतिक पैन कार्बोनाइजेशन सीएफ मैट्रिक्स के भीतर Bi_2O_3 और SnO_2 नैनोकणों के इन-सीटू न्यूक्लियेशन की सुविधा प्रदान करता है। एक अत्यधिक छिद्रपूर्ण 1डी संरचना जिसकी विशेषता मेसोपोरस छिद्र आकार और एक उच्च विशिष्ट सतह क्षेत्र है, इलेक्ट्रोलाइट की पहुंच को बढ़ाती है, तेजी से इलेक्ट्रॉन परिवहन को बढ़ावा देती है, और सुपरकैपेसिटिव प्रदर्शन के लिए महत्वपूर्ण प्रभावी चार्ज भंडारण तंत्र की सुविधा प्रदान करती है। यह अध्याय BS/NCFs के संरचना-संपत्ति संबंधों पर जोर देता है, जो उन्नत विद्युत रासायनिक गुणों को प्राप्त करने में इष्टतम एमओ संरचना और कैल्सीनेशन तापमान के लाभों पर प्रकाश डालता है।

अध्याय 8 शोध प्रबंध की पराकाष्ठा के रूप में कार्य करता है, जो पूरे अध्ययन में की गई जांचों का एक व्यापक सारांश और निष्कर्ष प्रस्तुत करता है। यह अंतिम अध्याय प्रमुख निष्कर्षों को संश्लेषित करने, उनके प्रभावों पर चर्चा करने और कुशल एससी इलेक्ट्रोड सामग्री को आगे बढ़ाने के लिए भविष्य की संभावनाओं को रेखांकित करने में महत्वपूर्ण है।

TABLE OF CONTENTS

CERTIFICATE	i
ACKNOWLEDGMENTS	ii
ABSTRACT.....	iv
TABLE OF CONTENTS.....	x
LIST OF FIGURES	xiv
LIST OF TABLES.....	xxii
ABBREVIATIONS AND SYMBOLS	xxiv
Chapter 1. Introduction	1
1.1 Background	1
1.2 Fundamental Principles of Supercapacitor.....	3
1.3 Classification of Supercapacitors	3
1.3.1 Electrical double-layer capacitors	3
1.3.2 Pseudocapacitors.....	7
1.3.3 Hybrid Supercapacitors.....	8
1.4 Electrode Materials	8
1.4.1 Carbon-based Materials	9
1.4.2 Metal Oxide-based Electrode Materials	10s
1.4.3 Metal Oxide and Carbon-based Hybrid Materials	10
1.5 Synthesis Procedure	11
1.5.1 Electrospinning	12
1.5.2 Post treatments	12
1.6 Challenges and Gap Area	13
1.7 Objectives	14
1.8 Structure of Thesis	15
1.9 References	17
Chapter 2. Methodology	23
2.1 Synthesis Methodology	23
2.1.1 Synthesis of WO ₃ decorated on PAN nanofibers	23
2.1.2 Synthesis of WO ₃ anchored to modified CNFs	24
2.1.3 Synthesis of 1D WO ₃ /SnO ₂ @C Nanocomposites	25

2.1.4 Synthesis of 1D porous SnO ₂ @C Fibrous Nanostructures	25
2.1.5 Synthesis of 1D Bi ₂ O ₃ and SnO ₂ embedded in CNFs	26
2.2 Material Characterizations	27
2.2.1 Powder X-ray Diffraction	27
2.2.2 Fourier Transform Infrared Spectroscopy	28
2.2.3 Raman Spectroscopy	29
2.2.4 Energy-dispersive X-ray Spectroscopy	30
2.2.5 X-ray Photon Spectroscopy	31
2.2.6 Electron Microscopy	32
2.2.7 Thermal Analysis	33
2.2.8 Surface Area Analysis	35
2.3 Electrochemical Measurements	35
2.3.1 Cyclic Voltammetry	35
2.3.2 Galvanostatic Charge Discharge	36
2.3.3 Electrochemical Impedance Spectroscopy	37
2.4 Electrochemical Analysis	38
2.4.1 Three Electrode Analysis	38
2.4.2 Two Electrode Analysis	39
2.5 Parameters Involved in Determining Electrochemical Performance	39
2.5.1 Specific Capacitance (C _{sp})	39
2.5.2 Energy Density (E _d) and Power Density (P _d).....	39
2.5.3 Charge Storage Mechanism	40
2.6 References	41
Chapter 3. Uniquely Designed Tungsten Oxide Nanopetals Decorated Electrospun PAN Nanofiber as Flexible Supercapacitor	44
3.1 Abstract	44
3.2 Introduction	45
3.3 Results and Discussions	46
3.3.1 Physicochemical Characterizations	46
3.3.2 Electrochemical Analysis in Three-electrode Configuration.....	53
3.3.3 Charge-Storage Mechanism of W-PAN Composite	56

3.3.4 Electrochemical Analysis in Two-electrode Configuration.....	59
3.4 Conclusion	61
3.5 References	62
Chapter 4. Tungsten Oxide Anchored Modified Carbon Nanofibers for Advanced Hybrid Supercapacitor Electrode	68
4.1 Abstract	68
4.2 Introduction	69
4.3 Results and Discussions	70
4.3.1 Physicochemical Characterizations	70
4.3.2 Electrochemical Analysis in Three-electrode Configuration	73
4.3.3 Optimizing WO ₃ Growth on m-CNFs	75
4.3.4 Electrochemical Analysis in Two-electrode Configuration... ..	77
4.4 Conclusion	79
4.5 References	79
Chapter 5. Metal oxide (WO₃/SnO₂) Heterostructure Supported Carbon Nanofibers for High Performance Supercapacitor Devices	83
5.1 Abstract	83
5.2 Introduction	84
5.3 Results and Discussions	85
5.3.1 Physicochemical Characterizations	85
5.3.2 Electrochemical Analysis in Three-electrode Configuration... ..	92
5.3.3 Electrochemical Analysis in Two-electrode Configuration... ..	96
5.4 Conclusion	99
5.5 References	100
Chapter 6. Highly Porous One-dimensional SnO₂-based Nanofibers for High Energy Density and High-Rate Capability Supercapacitors	107
6.1 Abstract	107
6.2 Introduction	108
6.3 Results and Discussions	109
6.3.1 Physicochemical Characterizations	109
6.3.2 Electrochemical Analysis in Three-electrode Configuration... ..	115
6.3.3 Electrochemical Analysis in Two-electrode Configuration... ..	117

6.4 Conclusion	119
6.5 References	119
Chapter 7. N-doped CNFs embedded with Bi₂O₃/SnO₂ for Enhanced Pseudocapacitive Energy Storage	122
7.1 Abstract	122
7.2 Introduction	123
7.3 Results and Discussions	124
7.3.1 Physicochemical Characterizations	124
7.3.2 Electrochemical Analysis in Three-electrode Configuration... ..	127
7.3.3 Discussion of metal precursor ratio and carbonization temperature	130
7.3.4 Charge Storage Mechanism for NCFs and BS/NCFs-10	131
7.3.5 Electrochemical Analysis in Two-electrode Configuration... ..	133
7.4 Conclusion	134
7.5 References	135
Chapter 8. Conclusion and Future Perspectives	141
8.1 Conclusion of Thesis	141
8.2 Future Perspectives	144
8.3 References	146
Appendices	147
Appendix 1	147
Appendix 2	155
Appendix 3	161
Appendix 4	171
Appendix 5	173
Curriculum Vitae	180

LIST OF FIGURES

Figure No.	Figure Caption	Page No.
1.1	The Ragone Plot compares energy density and power density between various electrochemical energy storage devices.	1
1.2	The components of a supercapacitor.	3
1.3	Schematic showing types of Supercapacitors according to their charge storage mechanism (a) EDLC (b) Pseudo capacitor.	4
1.4	a) Helmholtz, (b) Gouy-Chapman, and (c) Stern model of the electrical double-layer formed at a positively charged electrode in an aqueous electrolyte. The electrical potential, ϕ , decreases when transitioning from the electrode, ϕ_e , to the bulk electrolyte infinite away from the electrode surface, ϕ_s . The Stern plane marks the distance of closest approach of the ions to the charged surface. Note the absence of charges/ions in the Stern layer. The diffuse layer starts in the range of 10 – 100 nm from the electrode surface.	5
1.5	A schematic showing the double layer principle while (a) charging and (b) discharging of the electrode.	6
1.6	Taxonomy of supercapacitors.	8
1.7	(a) Pie-chart showing the data for the global SC market based on their type, Source: DataM Intelligence 2019. (b) The C_{sp} value for the materials being used supercapacitor electrodes for different types of SCs.	11
1.8	A typical schematic of the Electrospinning setup	12
1.9	Schematic showing preparation of metal oxide and carbon nanofiber-based composites using electrospinning followed by diverse post-treatment.	14
Scheme 2.1	The schematic illustration shows the synthesis for the W-PAN sample.	23
Scheme 2.2	The schematic illustration shows the synthesis for the WO_3/m -CNFs.	24
Scheme 2.3	An illustration of two-step protocol used for the synthesis of $WO_3/SnO_2@C$ fibers.	25
Scheme 2.4	Schematic representation of a flow chart for synthesizing $SnO_2@C$ via electrospinning followed by calcination at two different temperatures.	26
Scheme 2.5	Illustration of the synthetic process employed for the preparation of BS/NCFs.	27
2.1	(a) X-ray diffraction experiment, (b) Geometric derivation of Bragg's law: Constructive interference occurs when the delay between waves scattered from adjacent lattice planes given by $a_1 + a_2$ is an integer multiple of the wavelength λ .	28
2.2	The principle of operation and design of FTIR, (b) An image of the FTIR spectrophotometer at the Instrumentation Facility, Department of Chemistry, IIT Delhi, used in this thesis.	29

2.3	Schematic showing the Rayleigh scattering versus Stokes Raman and anti-Stokes Raman scattering, (b) A Picture of the Raman spectrometer at Central Research Facility (CRF, IITD) used for Raman analysis in this thesis.	30
2.4	Schematic showing the working principle of EDS. (b) A Picture of FESEM on a TESCAN Mira SEM equipped with an EDS detector at the Central Research Facility (CRF, IITD) used in this thesis.	31
2.5	(a) Schematic illustration of the photoemission process involved in XPS surface analysis. (b) A Picture PHI 5000 Versa Probe III instrument and Kratos Analytical limited (Axis Supra), IITD, used in this thesis.	32
2.6	(a) A Picture of the TEM instrument (JEOL 1400 microscope operating at 120 kV), (b) A Picture of High-Resolution Transmission Electron Microscopy (HR-TEM) (FEI Tecnai 20 operating at 200 kV), IITD used for imaging in this thesis.	33
2.7	(a) A typical TGA thermogram depicting the characteristic weight losses of electrospun polymer fibers, (b) A photograph of the TGA and DSC instruments used for thermal analysis in this thesis, conducted with the PerkinElmer instrument (Pyris Diamond, IITD).	33
2.8	Figure 2.8: (a)The five types of adsorption isotherms described by Brunauer, (b) A typical BET isotherm depicting the nitrogen adsorption-desorption analysis, inset: pore size distribution via BJH analysis for carbon nanofibers.	35
2.9	(a) Potential-time profiles used to perform cyclic voltammetry. (b) Sketch of a cyclic voltammogram of a pseudocapacitor showing both an EDLC response (red arrow, current proportional to scan rate) and an additional contribution of charge storage (blue arrows) through a redox couple with a defined apparent standard potential (E^0).	36
2.10	Schematic showing electrochemical configurations in (a) Three-electrode; (b) two-electrode	37
3.1	The FESEM images of (a) Electrospun PAN fibers (b) W-PAN sample.	47
3.2	The TEM images of W-PAN at two different locations in a TEM grid.	48
3.3	The elemental mapping images of the W-PAN sample, (a) without color coding, (b) mixed composite with bright colored spots, (c) tungsten, (d) oxygen, (e) carbon, and (f) nitrogen.	48
3.4	The X-ray diffraction pattern of the (a) W-PAN, (b) $WO_3 \cdot H_2O$, (c) electrospun pristine PAN fiber.	49
3.5	(Left panel) Raman spectra of (a) W-PAN sample; (b) $WO_3 \cdot H_2O$. (Right panel) FTIR spectra of (a) W-PAN; (b) $WO_3 \cdot H_2O$; (c) electrospun PAN fibers.	50
3.6	XPS spectra of the W-PAN Sample, showing high-resolution elemental scan of (a) W 4f, (b) Oxygen-1s, (c) Carbon-1s, (d) Nitrogen -1s.	52
3.7	CV curves at (a) different potential window, and (b) different scan rate for W-PAN. GCD curves at (c) different potential window, and (d) different current densities for W-PAN. (e) Ragone plot with inset; Specific capacitance of W-PAN as a function of current density. (f) Capacitance retention of W-PAN sample up to 1000 cycles at $40 A g^{-1}$.	54
3.8	Comparison between the $WO_3 \cdot H_2O$ and W-PAN; (a) CV at $100 mVs^{-1}$; (b) GCD at $10 A g^{-1}$, (c) Nyquist Plot at OCP (d) Specific capacitance comparison for $WO_3 \cdot H_2O$ and W-PAN.	55

3.9	(a) The trend of specific capacitance vs scan rate, (b) current response (anodic current) plotted against scan rate at different potentials, (c) b values plotted against voltages in the anodic voltammetric scan, (d) $i/v^{0.5}$ vs $v^{0.5}$ plot (e) capacitive contribution to charge storage at scan rates of 10 mV s^{-1} , (f) the percentage of capacitance contribution at different scan rates, (g) Inverse of stored charge ($1/C$) versus $v^{0.5}$, (stored charge (C) versus inverse of square root of the scan rate ($v^{-0.5}$).	58
3.10	Symmetric device W-PAN study, (a) CV curves at different potential windows, (b) CV curves at different scan rates, (c) GCD curves at different potential windows, (d) GCD curves at different current densities, (e) Capacitance retention with inset showing the cycle stability graph with cycle number, (f) Ragone Plot depicting energy density and power density with inset of specific capacitance of as-fabricated device as a function of current density.	60
4.1	(a-c) FESEM, (b) PXRD, (c) Raman, (c) FTIR for $\text{WO}_3/\text{m-CNFs}$, WO_3/CNFs , and WO_3 .	71
4.2	XPS CL spectra (a) W-4f, (b) O-1s, (c) C-1s, BET isotherm: Inset BJH Plot for WO_3 , WO_3/CNFs , and $\text{WO}_3/\text{m-CNFs}$.	73
4.3	Electrochemical comparison of WO_3 , WO_3/CNFs , and $\text{WO}_3/\text{m-CNFs}$ via (a) CV, (b) GCD, (c) b values calculated via $\log I$ vs \log scan rate(v) plot, (d) bar graph showing the capacitive and diffusive contributions extracting from Dunn's model at 10 mVs^{-1} , Three electrode characteristics of $\text{WO}_3/\text{m-CNFs}$ via (e) inverse specific capacitance ($1/C_{\text{sp}}$) as a function of square root of scan rate($v^{-1/2}$), with the y-intercept representing the total charge, (f) C_{sp} as a function of the inverse square root of scan rate($v^{1/2}$), with the y-intercept representing the surface charge.	75
4.4	FESEM images depicting the morphological evolution of $\text{WO}_3/\text{m-CNFs}$ hybrid electrodes with varying PAN to PVP ratios: (a) 1:0.15, (b) 1:0.25, (c) 1:0.5, and (d) 1:0.75, (d) EDX and (e) XPS survey spectra illustrating the atomic percent of Tungsten relative to carbon and oxygen content.	76
4.5	Electrochemical performance comparison of for $\text{WO}_3/\text{m-CNFs}$ with different PVP concentrations via (a) GCD at a constant current density of 8 A g^{-1} , (b) CV at varying scan rates, (c) b values calculated via $\log I$ vs \log scan rate(v) plot, (d) bar graph showing the capacitive and diffusive contributions extracting from Dunn's model at 10 mVs^{-1} .	77
4.6	Electrochemical comparison of $\text{WO}_3/\text{m-CNFs}$ // $\text{WO}_3/\text{m-CNFs}$ symmetric device via (a) CV, (b) GCD, (c) C_{sp} as function of current density, Ragone Plot, (d) Cyclic stability. Maximum potential window obtained by combing devices in series via (e) CV, and (f) GCD.	78
5.1	(a, b) XRD patterns, and (c) Raman and (d) FTIR spectra of $\text{SnO}_2@\text{C}$, $\text{WO}_3@\text{C}$, and $\text{WO}_3/\text{SnO}_2@\text{C}$. (b) enlarged portion of (a) showing XRD pattern in the 2θ range of 20° - 35° .	86
5.2	FESEM images of (a-c) electrospun and (d-f) calcined fibres of $\text{SnO}_2@\text{C}$, $\text{WO}_3@\text{C}$, and $\text{WO}_3/\text{SnO}_2@\text{C}$.	88
5.3	TEM and HRTEM (upper panel) images showcasing the morphology, and lattice fringes, respectively, at different grid locations and corresponding EDS elemental mappings (lower panel) for $\text{WO}_3/\text{SnO}_2@\text{C}$.	89

5.4	(a) Survey XPS spectra and high-resolution, XPS CL spectra for (b) W 4f; (c) Sn 3d; and (d) O 1s for WO ₃ /SnO ₂ @C composite, WO ₃ , and SnO ₂ fibers.	91
5.5	Comparative electrochemical response of the crafted samples via (a) CV curves at 100 mV s ⁻¹ , (b) GCD curves at 10 A g ⁻¹ , (c) CVs at varying scan rates, (d) GCD curves at varying current densities, (e) EIS within 1-10 ⁴ Hz, and (f) SSA normalized C _a , C _{sp} and pore size.	93
5.6	(a) anodic current versus scan rate (v) at varying potentials, (b) ‘b’ values plotted against voltages in anodic voltammetric scan and relative contribution (in %) at (c) varying scan rates and (d) at 9 mV s ⁻¹ for WO ₃ /SnO ₂ @C.	95
5.7	Response from symmetric full-cell device. (a, b) CVs at varying potential windows and scan rates, (c, d) GCD curves at varying potential windows and current densities, (e) Ragone plot with an inset showing C _{sp} at varying current densities, and (f) cyclic stability and coulombic efficiency for 1000 cycles at 2 A g ⁻¹ with an inset showing GCD curves at initial and final cycles.	96
5.8	(a) Pictures and (b) CVs showing the flexibility of the assembled device at various bending angles. (c) A schematic showing the real-time working of the assembled device and (d) lighting of Red LED using three symmetric devices stacked in series.	97
6.1	X-ray Diffraction (XRD) Pattern (b) FT-IR of SnO ₂ annealed at 400°C and 600°C.	110
6.2	SEM images and corresponding size distribution histograms of the as-synthesized electrospun nanofibers; (a) and (b) PVP -SnO ₂ composite, (c) and (d) SnO ₂ -400, and (e) and (f) SnO ₂ -600.	111
6.3	(a–e) Elemental mapping of calcined electrospun SnO ₂ -600 for different elements Sn, O, and C. (f) The SEM-EDX spectrum.	112
6.4	Figure 6.4: (a) TEM images for SnO ₂ nanofibers annealed at 600 °C, (b) SAED pattern of an SnO ₂ -600 fibre and (c) Raman spectrum of the SnO ₂ nanofibers annealed at 600 °C.	113
6.5	XPS analysis of SnO ₂ -600 sample. (a) Full survey scan and high-resolution core level XPS spectra of individual elements for (b) tin Sn 3d, (c) carbon C 1s, and (d) oxygen O 1s.	114
6.6	CV curves were recorded with a carbon cloth working electrode to study the effect of (a) catalyst material, (b) potential window, and (c) scan rate on the capacitive properties. GCD profile at (d) varying potential window and (e) varying current densities, (f) Variation of C _{sp} with current densities, (g) cycle stability for 1000 cycles, (h) Ragone plot for E _d and P _d , and (i) schematic diagram for electrochemical reaction.	113
6.7	(a) CV plots for both SnO ₂ @C-600 and activated carbon. CV curves for ASC at (b) various potential windows and (c) different scan rates. (d) GCD profile at various current densities. (e) Cyclic stability and Variation of specific capacitance with current densities(inset) and (f) Ragone Plot (E _d Vs P _d) of the ASC device.	118
7.1	(a) XRD, (b) FTIR, (c) Gaussian fitting of FTIR peaks in the range of 1400–1800 cm ⁻¹ , and (d) Raman for NCFs, and BS/NCFs samples.	125

7.2	(a-d) FESEM images and corresponding size distribution histograms of the as-synthesized samples (a, e) NCFs, (b, f) BS/NCFs-2, (c, g) BS/NCFs-5, (d, f) BS/NCFs-10.	126
7.3	Deconvoluted Core Level (CL) spectra of (a) C-1s, (b) N-1s, (d)O-1s, (e) Bi 4f, (f) Sn-3d, (c) Distribution of N sites for NCFs, and BS/NCFs samples.	128
7.4	Electrochemical performance for NCFs, and BS/NCFs samples via (a) comparative CVs at 25 mV s ⁻¹ , (b) CV at varying scan rate, (c) C _{sp} estimated from CV at varying scan rate (d) Comparative ECSA calculation, (e) C _{sp} estimated from CV at varying current densities, (f) capacitance retention up to 5000 cycles.	129
7.5	Elucidation of Pseudo capacitance and EDLC contribution in the total C _{sp} for NCFs and BS/NCFs-10 via (a) Discharge time value measured at 80 A g ⁻¹ , (b) inverse specific capacitance (1/C _{sp}) as a function of square root of scan rate(v ^{-1/2}), with the y-intercept representing the total charge, (c) C _{sp} as a function of inverse square root of scan rate(v ^{1/2}), with the y-intercept representing the surface charge. (d) b values calculated through the experimental CV via log I vs log scan rate(v) plot, (e, f) CV showing the capacitive and diffusive contributions extracting from Dunn's model at 10 mVs ⁻¹ .	132
7.6	Electrochemical evaluation of BS/NCFs-10 in a two-electrode system using PVA/H ₂ SO ₄ gel electrolyte: CV curves recorded (a) at various scan rates in the fixed potential window of 0.0–1.1 V. (b) fixed scan rate of 100 mV s ⁻¹ in various potential windows ranging from 0.0 to 1.4 V. Charge/discharge curves at (e) different current densities in the potential window of 0.0–1.1 V. (f) different potential window of 0.0–1.1 V at a constant current density. (g) Ragone Plot: Inset C _s versus current density. (h) Capacitance retention as a function of cycle number.	134
8.1	Schematic illustration depicting the performance evaluation of MO/polymer hybrid electrodes explored in the thesis.	142
8.2	Schematic illustration of <i>in situ/operando</i> spectroscopic techniques for charge storage mechanism	145
A1.1	SEM images of tungsten oxide, PAN, and W-PAN	147
A1.2	(a) Particle size distribution of the PAN fibers. (b) and (c) FESEM images of PAN fiber.	148
A1.3	(a) The elemental mapping images of PAN sample (b) mixed composite with bright coloured spots, (c) carbon, (d) nitrogen, (e) oxygen, and (g) the corresponding EDX spectrum	148
A1.4	The corresponding the EDX spectrum for W-PAN sample.	149
A1.5	The thermal gravimetric analysis of PAN fiber and W-PAN sample.	149
A1.6	The BET surface area and pore size distribution graphs of pristine WO ₃ .H ₂ O and W-PAN samples.	150
A1.7	XPS spectra of the W-PAN Sample, depicting the survey scan	150
A1.8	XPS spectra of the PAN Sample high-resolution elemental scan of (a) Carbon-1s; (b) Nitrogen-1s; (c) Oxygen-1, (d) comparison of O-1s of W-PAN, and WO ₃ .H ₂ O and XPS comparison of C-1s of W-PAN, and PAN.	151
A1.9	(a) Cyclic voltammetry comparison of W-PAN in the potential range of -0.25 to 0.2V and (b) GCD curves at 6 A g ⁻¹ .	152
A1.10	Nyquist plots of the W-PAN electrode.	152

A1.11	CV comparison of the W-PAN, PAN, and pristine WO ₃ .H ₂ O at scan rate 25 mVs ⁻¹	153
A1.12	Nyquist plots of the symmetric device; inset shows the magnified high-frequency region.	153
A 2.1	(a) FESEM image and (b) corresponding EDX spectrum of WO ₃ /m-CNFs (0.5). EDX mapping depicting (c) the mixed elemental composition, and individual elements (d - f) W, O, and Cor WO ₃ /m-CNFs(0.5).	155
A2.2	(a) FESEM image and (b) corresponding EDX spectrum of WO ₃ /m-CNFs (0.5). EDX mapping depicting individual elements (c - f) W, O, C and N for WO ₃ /CNFs.	155
A2.3	(a)XRD pattern for m-CNFs showing characteristic (002) peak, (b)Enlarged view of the (a) XRD pattern in the 2θ range of 20 ⁰ -35 ⁰ , (c) FTIR in the wavenumber range of 1400-600 cm ⁻¹ for WO ₃ , WO ₃ /CNFs, and WO ₃ /m-CNFs.	156
A2.4	Comparative (a)FTIR analysis, (b) Raman spectra for CNFs and m-CNFs samples.	157
A2.5	Comparative XPS CL C-1s spectra for CNFs and m-CNFs samples	157
A2.6	Three electrode characteristics of WO ₃ /m-CNFs via CV at different scan rates, Electrochemical comparison of WO ₃ , WO ₃ /CNFs, and WO ₃ /m-CNFs via (b) Cyclic stability at 10 A g ⁻¹ and (c) EIS in the frequency range of 1 to 10 ⁴ Hz.	158
A2.7	Electrochemical performance of WO ₃ /m-CNFs (0.5) via (a)CV at constant scan rate of 25 mVs ⁻¹ , (b) GCD at constant current density of 8 A g ⁻¹ at varying potential window, (c) GCD at varying current density (c) Energy density and power density in Ragone Plot, Inset: C _{sp} as a function of current.	159
A2.8	Electrochemical performance of WO ₃ /m-CNFs (0.5) //WO ₃ /m-CNFs (0.5) symmetric device at varying potential window via (a) CV at constant scan rate of 100mVs ⁻¹ , (b) GCD at 4 Ag ⁻¹	159
A2.9	EIS comparison of WO ₃ /m-CNFs (0.5) //WO ₃ /m-CNFs (0.5) before and after cyclic stability in the frequency range of 1 to 10 ⁴ Hz.	160
A2.10	CV comparison of WO ₃ /m-CNFs (0.5) //WO ₃ /m-CNFs (0.5) before and after cyclic stability in the potential range of 0 to 1.4 V at 100 mVs ⁻¹ .	160
A3.1	Zoomed view of Raman spectra for SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers showing (a) M-O bonds in the range of 500-850 cm ⁻¹ , (b) Deconvoluted Raman Spectra using Voight Function for (c) WO ₃ /SnO ₂ @, (c) FTIR spectra in the wavenumber range of 1500-400 cm ⁻¹ (left panel) for SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers.	161
A3.2	The size distribution histograms of the (a-c) as-synthesized electrospun nanofibers of SnO ₂ , WO ₃ and WO ₃ /SnO ₂ , (d-e) Calcined Fibers of SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers.	162
A3.3	a) FESEM image and (b) corresponding EDX spectrum of WO ₃ /SnO ₂ @C fibers. EDX mapping depicting (c) the mixed elemental composition, and individual elements (d - j) W, Sn, O, C, and N for WO ₃ / SnO ₂ fibers.	162
A3.4	(a) STM images depicting the area of interest for EDS analysis, (b) The corresponding elemental overlap of WO ₃ /SnO ₂ @C fibers.	163
A3.5	(left) SAED pattern recorded for WO ₃ /SnO ₂ @C fibers. (right) The thermogravimetric analysis of the SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers.	163

A3.6	High-resolution XPS CL spectra (a) C 1s for WO ₃ /SnO ₂ @C, WO ₃ @C, SnO ₂ @C, and PAN fibers, Before and after the cycling stability (b) W-4f, (c) Sn 3d for WO ₃ /SnO ₂ @C.	164
A3.7	(a) The N ₂ adsorption-desorption analysis, (b) pore size distribution via BJH analysis of the SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers.	164
A3.8	Left panel: The comparative CV curves of WO ₃ / SnO ₂ @C fibers in the potential range of 0–1 V at a scan rate of 100 mV s ⁻¹ in different electrolytes. Right Panel: The CV curve of WO ₃ @C and SnO ₂ @C fibers in the potential range of 0–0.8 V at a scan rate of 25 mV s ⁻¹	165
A3.9	Electrochemical performance of WO ₃ /SnO ₂ @C (a) via CVs, (b) GCD at varying potential windows, (c) Variation of C _{SP} (from GCD analysis) with current density, and (d) corresponding Ragone plot.	165
A3.10	(a) The Bode phase angle plot, (b) stability stability test till 5000 cycles at 10 A g ⁻¹ for WO ₃ @C, SnO ₂ @C, and WO ₃ /SnO ₂ @C fibers.	166
A3.11	Comparison of PAN, WO ₃ / SnO ₂ @C, via (a) CV in the potential range of 0 to 0.9 V at 25 mVs ⁻¹ ; (b) GCD in the potential range of 0 to 0.9 V at 10 Ag ⁻¹ ; (c) cyclic stability up to 5000 cycles at 10 A g ⁻¹ ; (d) EIS in the frequency range 1- 10 ⁴ Hz at OCP: Inset Zoomed EIS of PAN with Equivalent circuit elements	166
A3.12	FESEM of WO ₃ /SnO ₂ @C before and after 5000 GCD cycles.	167
A3.13	A comparison of Cdl values for SnO ₂ @C, WO ₃ @C, and WO ₃ /SnO ₂ @C fibers.	167
A3.14	CV with current density normalized by SSA, and Ca values for WO ₃ @C, SnO ₂ @C, and WO ₃ /SnO ₂ @C fibers	168
A3.15	(a) anodic current density versus square root scan rate (v ^{1/2}) at peak potentials, (b) I/v ^{0.5} vs v ^{0.5} plot for WO ₃ /SnO ₂ @C, the percentage of capacitance contribution at different scan rates for (c) WO ₃ @C, (d) SnO ₂ @C.	169
A3.16	Nyquist plot for the symmetric device (WO ₃ /SnO ₂ @C) in the frequency range 0.1 to 10 ⁴ Hz	170
A4.1	(a) The N ₂ adsorption-desorption analysis, Inset: pore size distribution via BJH analysis of the SnO ₂ -600.	172
A4.2	EIS spectroscopy of SnO ₂ with fitted circuit diagram, high-frequency region (inset).	172
A5.1	(a) TGA, (b) Differential TGA of NCFs and BS/NCFs samples.	173
A5.2	TEM images BS/NCFs-10 at different locations in a TEM grid.	175
A5.3	(a) FESEM image and (b) corresponding EDX spectrum of BS/NCFs-10, EDX mapping depicting (c) the mixed elemental composition, and individual elements (d - h) Bi, Sn, O, C, and N for BS/NCFs-10.	175
A5.4	BET isotherm plot (a) N ₂ adsorption-desorption and (b) BJH pore size distribution for the NCFs and BS/NCFs-10 sample.	176
A5.5	Comparative (a) C _{sp} estimated from CV at varying scan rate (d) ECSA calculation, Electrochemical performance via (a) CV at a constant scan rate of 50 mVs ⁻¹ and different applied potential window; (b) GCD at different current densities at the constant applied potential window; (c) Specific capacitance at different applied current density. (d) The Ragone plot with Energy density (E _d) and Power density (P _d) for NCFs and BS/NCFs sample.	176
A5.6	(a,d) FESEM images and Deconvoluted Core Level (CL) spectra of (b,e)	177

	C-1s;(c,f) N-1sfor BS/NCFs-80, and BS/NCFs-10(600) samples.	
A5.7	Charge storage mechanism in BS/NCFs-10: (a) (c) b values calculated through the experimental CV via log I vs log scan rate(v) plot, (b) plot of $(I)/v^{1/2}$ vs $v^{1/2}$.	178
A5.8	The percentage of capacitance contributions at different scan rates for (a) NCFs, (b) BS/NCFs-2, (c) BS/NCFs-5, (d) BS/NCFs-10; CV showing the capacitive and diffusive contributions extracted from Dunn's model at 10 mVs^{-1} for (e) BS/NCFs-2, (f) BS/NCFs-5.	178
A5.9	(a) EIS recorded at OCP in the frequency range of 0.01 Hz to 100 kHz; the inset shows the Schematic of the symmetric SC; CV curves symmetric SC recorded at a (b) fixed scan rate of 100 $mV s^{-1}$ before and after cyclic stability; (c) fixed scan rate of 100 $mV s^{-1}$ at different bending angles.	179

LIST OF TABLES

Table No.	Table Caption	Page No.
1.1	<i>Performance Comparison Between Batteries, Capacitors, and Supercapacitors.</i>	2
3.1	<i>Comparison of electrochemical performances of various carbon-based Tungsten oxide electrode materials.</i>	61
5.1	<i>The peak positions of C, O, Sn, and W in SnO₂@C, WO₃@C, and WO₃/SnO₂@C were obtained from XPS analysis.</i>	91
5.2	<i>Comparison of electrochemical parameters, viz., voltage drop, discharge time, ESR, C_{SP}, and capacitance retention obtained from half-cell configuration for the studied samples.</i>	93
5.3	<i>Comparison of electrochemical performance parameters for WO₃-based electrodes reported in the literature.</i>	98
7.1	<i>Comparison of super capacitive performance parameters for various Metal oxide embedded CNFs electrodes reported in the literature.</i>	130
A1.1	<i>EDX table of W-PAN exhibiting the atomic % of the individual elements</i>	153
A1.2	<i>EDX table of PAN exhibiting the atomic % of the individual element</i>	154
A2.1	<i>A summary of Trasatti analysis exhibiting the C_{total}, C_{inner}, and C_{outer} for WO₃, WO₃/CNFs, and WO₃/m-CNFs samples.</i>	158
A3.1	<i>Details of the BET surface areas and average pore diameter of MO@C fibers calculated from the N₂ adsorption-desorption isotherms</i>	162
A3.1	<i>Details of the electrochemical performance of WO₃/SnO₂@C in half configuration</i>	168
A3.5	<i>Details of the R_s and Warburg coefficient (σ) of MO@C fibers calculated using the Randles circuit.</i>	169
A5.1	<i>Results of I_D/I_G for NCFs and BS/NCFs</i>	173
A5.2	<i>Results of EDX, ICP-OES, and XPS showing atomic % for NCFs and BC/NCFs samples.</i>	174
A5.3	<i>The relative area percentages for NCFs and BS/NCFs sample as determined from XPS</i>	174

ABBREVIATIONS AND SYMBOLS

ASC	Asymmetric Supercapacitor
BJH	Barrett-Joyner-Halenda
B.E.	Binding Energy
BET	Brunauer–Emmett–Teller
CNFs	Carbon nanofibers
CL	Core Level
CV	Cyclic voltammetry
DTA	Differential Thermal Analysis
EDLC	Electrochemical double-layer capacitance
EESS	Electrochemical energy storage systems
EIS	Electrochemical impedance spectroscopy
ES	Electrospinning
ESR	Equivalent Series Resistance
FESEM	Field Emission Scanning Electron Microscopy
FTIR	Fourier Transform Infrared Spectroscopy
GCD	Galvanostatic Charge Discharge
MO	Metal oxide
NS	Nanostructures
NCFs	Nitrogen-doped carbon fibers
1D	One dimensional
O _{vac}	Oxygen Vacancy
OCP	Open circuit potential
PVP	Polyvinylpyrrolidone
PAN	Polyacrylonitrile
SAED	Selected Area Electron Diffraction
STEM	Scanning Tunneling Microscopy
SEM	Scanning Electron Microscopy
SC	Supercapacitor
SSA	Specific Surface Area
TEM	Transmission Electron Microscopy
XRD	X-ray Diffraction
XPS	X-ray Photon Electron Microscopy
C _a	Areal Capacitance
E _d	Energy Density
P _d	Power Density
C _{sp}	Specific Capacitance
h	hour
min	minute
s	second
g	gram

eV	Electron volt
°C	degree centigrade
M	molar
R_{ct}	charge transfer resistance
R_s	solution resistance
σ	Warburg coefficient
t	time
V	voltage
V_{drop}	voltage drop/IR drop
ϵ_0	permittivity of vacuum
ϵ_r	relative dielectric constant