

**STUDIES ON CO₂-INDUCED CRYSTALLIZATION AND
FOAMING BEHAVIOUR OF POLY(LACTIC ACID)/CLAY
NANOCOMPOSITES**

SABAPATHY S



**DEPARTMENT OF MATERIALS SCIENCE AND ENGINEERING
INDIAN INSTITUTE OF TECHNOLOGY DELHI**

SEPTEMBER 2018

©Indian Institute of Technology Delhi (IITD), New Delhi, 2018

**STUDIES ON CO₂-INDUCED CRYSTALLIZATION AND
FOAMING BEHAVIOUR OF POLY(LACTIC ACID)/CLAY
NANOCOMPOSITES**

by

SABAPATHY S

Department of Materials Science and Engineering

Submitted

in fulfilment of the requirements of the degree of Doctor of Philosophy

to the



INDIAN INSTITUTE OF TECHNOLOGY DELHI

SEPTEMBER 2018

DEDICATED TO DIVINE NATURE

Certificate

This is to certify that the thesis entitled “studies on CO₂-induced crystallization and foaming behaviour of poly(lactic acid)/clay nanocomposites” being submitted by Mr. Sabapathy S to the Indian Institute of Technology Delhi, for the fulfilment of award of the degree “Doctor of Philosophy” is a record of bonafide research work carried by him under our supervision. This thesis has been prepared in conformity with the rules and regulations of the Indian Institute of Technology Delhi, New Delhi. We further certify that the thesis has attained a standard required for a Ph.D. degree of the institute. The research reported and results presented in the thesis have not been submitted in part or full to any other institute or university for the award of any other degree or diploma.



(Anup Kumar Ghosh)

Professor and Head
Department of Materials Science and Engineering,
(Formely Centre for Polymer Science & Engineering)
Indian Institute of Technology Delhi,
Hauz Khas, New Delhi -110016, India.



(Chul B. Park)

Distinguished Professor of Microcellular Engineered Plastics,
Department of Mechanical & Industrial Engineering,
University of Toronto,
M5S 3G8, 5 King's College Rd., Ontario, Canada

Acknowledgments

It gives me immense pleasure to express my deep sense of gratitude to all who have helped me along my way through the doctoral studies and a memorable stay at IIT Delhi

*I express my profound sense of gratitude and veneration to my supervisors, **Prof. Anup K. Ghosh and Prof. Chul B. Park** for guiding me all the way during my PhD tenure. Their constant inspiration and encouragement have been a great motivation for me behind all the work I have conducted. I appreciate their cool composure and patience which helped me to tide over the difficult situations.*

I express my deep sense of gratitude to my student research committee members Prof. S.N Maiti, Dr. Bhabani K. Satapathy, and Prof. Naresh Bhatnagar who have monitored my work and provided me the valuable suggestions. I am also thankful to Prof. Veena Choudhary, Prof. Jacob Josemon, Dr. Leena Nebhani, and Dr. Sampa Saha who inspite of their busy schedule have always made themselves available for valuable discussions and support.

I owe thanks to laboratory staffs Mr. Ashok Kapoor, Mr. Surender Sharma, Mr. Shiv Kant, Mr. Ehteshamul Islam and Mr. Gajraj Singh for their timely help and suggestions. I would also like to convey my thanks to Mr. Narender Kumar, Ms. Shalini Arora, Mr. Sudhir Kumar Pandey and Pramod Kale for their all possible supports. I am grateful to Mr. D. C. Sharma and Mr. Kuldeep Sharma of SEM central facility for teaching and allowing me the SEM of my samples.

I would like to extend my acknowledge my friends who has made my life colorful and enjoyable. I am grateful to Debanga konwar, Sampat Singh Chauhan, Vishvapratap Singh, Dr. Bhavana Sharma, Pragati and Banpreet,

I would like to extend my acknowledge to my research colleagues of Microcellular Polymer processing research group. Their friendships are integral parts of my graduate studies experience. Many of my research works would not have been as successful without their

advice and assistance. I am thankful to Mr. Anindya Dutta, Ms. Ritima Banerjee, Ms. Swarna. and Mr. Jung Hyub Lee.

I would like to convey my special thanks to Dr. Jutika Goswami, Dr. Sanjeev Kumar, Dr. Tahir Zafar, Dr. Manash, Dr. Priyanka Singh for keeping him/her available for me from time to time with her/his valuable suggestions and encouraging spirit.

Finally and above all, I would like to thank my parents and priya. It would not have been possible for me to reach at this stage of academic pursuit without their love and motivation.

A handwritten signature in black ink, appearing to read 'S. Sabapathy', with a long horizontal stroke extending to the right.

SABAPATHY S

ABSTRACT

The commercial grades poly(lactic acid) (PLA) are usually based on an L-rich mixture and typically comprise of 95-97 % L-units and 1-2% D-units. PLA exhibits very slow crystallizing behaviour. During foam processing of PLA with CO₂, PLA/CO₂ system exhibits rapid crystallization behaviour which imparts high stable crystalline foam morphology to the final product. This CO₂ induced crystallization process occur at supersaturated/foaming condition is spontaneous in nature. This indicates that like bubble nucleation, crystal nucleation in a CO₂ supersaturated samples is a spontaneous process and it can be enhanced by increasing the degree of supersaturation. The objective of the current thesis is to investigate the crystallization and foaming behaviours of PLA and PLA/clay nanocomposites (PLANCs) at their CO₂ induced supersaturation state for the development of structurally controlled crystalline foam morphologies. The raw material used were PDLA (PLA8032D), nanoclay (Cloisite 30B) and high pressure CO₂ as green plasticizer. PLA and Cloisite 30B were melt compounded in a Prism EUROLAB-16 co-rotating twin screw extruder. PLA based nanocomposites containing 0.5 wt.% (PLANC-0.5), 1 wt.% (PLANC-1), and 2 wt.% (PLANC-2) of nanoclay were prepared. The various characterizations of PLA/clay nanocomposites (PLANCs) were carried out to study the effects of nanoclay content (wt.%) on the thermal, morphological, rheological, mechanical properties and thermo-mechanical properties of the PLANCs. The crystallization behaviour of PLA and PLANCs were studied at saturation and supersaturation condition with CO₂. The distinct differences in the crystal growth behaviour between the two states were investigated. The crystallization behaviour of CO₂ treated PLA and PLANCs at various CO₂ saturation conditions (20, 30 & 50 bar) were performed by using high pressure differential scanning calorimetry (HP-DSC). The crystallization behaviour of CO₂ treated PLA and PLANCs under supersaturated conditions were studied by using high pressure autoclave. The compression moulded amorphous PLANCs samples were saturated

with CO₂ at an optimum conditions using high pressure autoclave. In order to achieve various degree of supersaturation, the CO₂ saturated samples were heated to different temperatures (60 °C, 70 °C, 75 °C, 80 °C). The resultant spherulitic structures were investigated by using optical microscopy, scanning electron microscopy, and wide angle X-ray diffraction. In contrast to CO₂ saturated conditions (50 bar), there was an exponential increases in nucleation density of spherulites (from 220 to 4.5×10^3 number/mm²) at superheated condition (80 °C) was observed. Interestingly, from the morphological analysis, it was established that the superheating process not only enhances nucleation exponentially but also induces transition from 3-D to 2-D crystal growth mechanism in PLA. To establish the correlation between the crystallization and foaming process, concurrent crystallization and foaming behaviour of PLANCs were studied at different superheated conditions (120 °C, 130 °C, 140 °C, and 150 °C). The obtained 2-D crystalline foam morphologies of PLANCs were studied by various characterization techniques such as, SEM, SAXS and WAXD techniques. It was found that the CO₂ induced 2-D crystal growth mechanism of PLANCs with exclusion of CO₂ simply direct the growth of 2-D spherulites into the self-templated regions and in-situly foam the inter-spherulitic amorphous region concurrently, so-called ‘instantaneous 2-D crystallization and in-situ foaming processes’. The obtained self-stack 2-D spherulites with in-situ foamed interlayers replicate the 2-D crystalline porous microstructure of nacre biomaterials and thus the rapid biomimetic concept of crystallization of PLANCs occurs at superheated conditions. The subsequent transformation of the amorphous PLA into a self-stacked 2-D spherulitic microstructure and the reinforcing effect of semi-crystalline 2-D spherulites in the novel assembly with in-situ foamed amorphous interface were established by various mechanical studies. This thesis reveals the rapid biomimetic concept of CO₂ induced crystallization and foaming of PLA & PLANCs occurring at superheated conditions and its application in developing bioinspired polymeric materials for structural applications.

Table of Contents

Certificate.....	i
Acknowledgements.....	ii
Abstract.....	iv
Table of Content.....	vi
List of Figures.....	xi
List of Tables.....	xix
List of Symbols.....	xxi
List of Abbreviations.....	xxii

CHAPTER 1: INTRODUCTION AND LITERATURE SURVEY

1.1 Poly(lactic acid) (PLA).....	1
1.2 Physical properties of PLA.....	2
1.2.1 Advantages of PLA.....	2
1.2.2 Limitations of PLA.....	3
1.3 PLA based bio-foams.....	3
1.3.1 Advantages of PLA based biofoams.....	3
1.3.2 Challenges in foam processzbility of biofoams.....	4
1.3.3 Development of foam morphology.....	4
1.3.4 Applications of PLA based bone scaffolds.....	5
1.4 PLA/clay nanocomposites foam.....	7
1.4.1 Preparation of PLA/clay nanocomposites.....	7
1.4.2 Properties of PLA/clay nanocomposites.....	8
1.4.3 Properties of PLA/clay nanocomposite foams.....	11
1.5 PLA crystallinity.....	14
1.5.1 Effect of D-content on PLA crystallization.....	16
1.5.2 Effect of plasticization on PLA crystallization.....	17
1.5.3 Effect of nucleation and plasticization on PLA crystallization.....	18
1.6 Crystallization process of biomaterials under supersaturation states.....	20
1.6.1 Nacre (Mother of pearl).....	21
1.6.2 Preparation of artificial nacre.....	24
1.6.3 Non-classical crystallization process.....	25

1.6.4	Biomimetic crystallization process	25
1.7	sc-CO ₂ assisted polymer processing	28
1.7.1	Solubility of CO ₂ in PLA	31
1.7.2	CO ₂ induced plasticization.....	34
1.7.3	CO ₂ induced crystallization process	35
1.7.4	CO ₂ induced foaming process.....	38
1.7.5	CO ₂ induced crystallization and foaming processes.....	41
1.7.6	CO ₂ -induced crystallization process at supersaturation states.....	43
1.8	Literature summary	46
1.9	Scope.....	48
1.10	Objective and plan of work.....	48
1.11	Format of the thesis.....	50

CHAPTER 2: MATERIALS AND EXPERIMENTAL METHODS

2.1	Introduction.....	61
2.2	Raw material characterization.....	62
2.2.1	PLA	62
2.2.1.1	Thermal characterization	63
2.2.1.2	Infrared spectroscopy.....	64
2.2.2	Nanoclay	65
2.2.2.1	XRD, TGA and SEM analysis.....	66
2.3	Preparation of PLA based clay nanocomposites (PLANCs)	68
2.4	Thermogravimetric Analysis (TGA).....	70
2.5	Differential Scanning Calorimetry (DSC)	70
2.6	Capillary rheometer	71
2.7	Parallel plate rheometer	72
2.8	Injection moulding.....	72
2.9	Transmission electron microscopy (TEM)	72
2.10	Mechanical properties.....	73
2.11	Dynamic mechanical analysis (DMA).....	73
2.12	Compression moulding	74
2.13	Gas solubility	74
2.14	Polarized optical microscopy (POM).....	75
2.14.1	Crystallization morphology of unplasticized PLANCs	75

2.14.2	Crystallization morphology of CO ₂ plasticized PLANCs	75
2.15	High pressure differential scanning calorimetry (HP-DSC).....	75
2.16	CO ₂ induced crystallization studies at superheated conditions.....	77
2.16.1	CO ₂ induced crystallization studies on PLANCs' films	77
2.16.2	CO ₂ induced crystallization studies on PLANCs' thick specimens.....	78
2.17	CO ₂ induced crystallization and foaming process	79
2.18	WAXD analysis	80
2.19	Scanning electron microscopy (SEM)	81
2.20	Specific density of foamed samples.....	82

CHAPTER 3: THERMAL, RHEOLOGICAL, MORPHOLOGICAL THERMO-MECHANICAL STUDIES OF PLA & PLA BASED CLAY NANOCOMPOSITES.

3.1	Introduction.....	85
3.2	Clay morphology analysis of PLANCs.....	85
3.3	Dynamic rheology.....	89
3.4	Thermal behaviour	92
3.5	Thermogravimetric analysis.....	95
3.6	DMA analysis	97
3.7	Shear flow rheology	99
3.8	Mechanical properties of PLANCs	101
3.8.1	Tensile properties.....	101
3.8.2	Impact properties	102
3.9	Summary	103

CHAPTER 4: STUDIES ON CRYSTALLIZATION BEHAVIOR OF CO₂ PLASTICIZED PLA AND PLA/CLAY NANOCOMPOSITES AT EQUILIBRIUM CONDITIONS

4.1	Introduction.....	106
4.2	Isothermal melt crystallization kinetics of PLANCs.....	107
4.2.1	Spherulite morphology studies of PLANC's	114
4.3	Isothermal melt crystallization kinetics of CO ₂ plasticized PLA	116
4.4	Isothermal crystallization kinetics of CO ₂ plasticized PLANCs.....	122
4.4.1	Crystalline morphology of isothermally crystallized PLANCs.....	127
4.5	Summary.....	130

CHAPTER 5: STUDIES ON CO₂-INDUCED CRYSTALLIZATION BEHAVIOUR OF PLA AND PLA/CLAY NANOCOMPOSITES AT SUPERHEATED CONDITIONS

Chapter 5A: CO ₂ induced crystallization of PLA and PLA/clay nanocomposites films at superheated conditions.....	133
5A.1. CO ₂ saturation process of PLA films	134
5A.2. Gas solubility of PLANCs.....	135
5A.3. Classical description of crystal nucleation of a polymer in supersaturated conditions	136
5A.4. In-situ observation of CO ₂ -induced crystallization process of PLA	139
5A.5. Optical microscopy.....	139
5A.6. WAXD analysis of CO ₂ induced crystallized PLA films.....	143
5A.7. SEM analysis of CO ₂ induced 2-D crystallized PLA films.....	144
5A.8. Optical micrographs of PLANCs	145
5A.9. Summary.....	147
Chapter 5B: Morphological analysis of CO ₂ induced crystallized PLA and PLA/clay nanocomposites (thick moulded samples) at superheated conditions.....	149
5B.1 The CO ₂ -induced crystallization process	150
5B.2 Crystalline morphological techniques of PLA	151
5B.3 Scanning electron microscopy analysis.....	152
5B.4 Optical profilometry	155
5B.5 WAXD analysis.....	157
5B.6 Optical microscopy analysis.....	159
5B.7 Effect of nanoclay on CO ₂ -induced 2-D crystalline microstructure	162
5B.8 Crystal nucleation and growth mechanism of polymer in supersaturated conditions	164
5B.9 Biomimetic approach of crystallization	165
5B.10 Summary	166

CHAPTER 6: STUDIES ON CO₂ INDUCED FOAM PROCESSABILITY OF PLA AND PLA/CLAY NANOCOMPOSITES.

Chapter 6A: CO ₂ induced crystalline and foam morphologies of PLA and PLA/clay nanocomposites at superheated conditions.	169
6A.1 Effect of degassing temperature on the spherulites microstructure of PLA-NCF	171
6A.2 Crystalline and foam morphological analysis of PLA-NCF	172
6A.3 Effect of superheat on the brick-and-mortar microstructure.....	176

6A.4 Effect of nanoclay on crystalline foamed PLANC-Cs.....	181
6A.5 Crystal structure and degree of crystallinity of PLANC-CFs	184
6A.6 Self-association of clay platelets in 2-D spherulites	185
6A.7 Biomimetic crystallization	187
6A.8 Summary	191
Chapter 6B: Mechanical properties and structural co-relationship of structurally controlled crystalline foamed PLA and PLA/clay nanocomposites	193
6B.1. Impact properties of “Stack of coins” and “brick-and-mortar” crystalline microstructures	194
6B.2. Bending properties of CO ₂ -induced crystallize foamed PLA samples (PLA-NCF) under static loading conditions.....	198
6B.3. Bending properties of PLA-NCF and PLANC-CFs at dynamic loading conditions.....	205
6B.4. Flexural strength of CO ₂ induced PLANC-CFs	210
6B.5. Summary.....	211

CHAPTER 7: SUMMARY AND CONCLUSIONS

7.1 Summary.....	217
7.1.1. Melt compounding of the PLA based claynanocomposites and their characterization.....	217
7.1.2. Crystallization studies of CO ₂ plasticized PLANCs at saturated conditions	219
7.1.3. CO ₂ -induced crystallization studies of PLA and PLANCs at superheated conditions	220
7.1.4. Morphological analysis of CO ₂ -induced 2-D crystallized PLA and PLANCs	221
7.1.5. CO ₂ induced crystallization and foaming studies of PLA and PLANCs.....	222
7.1.6. Comparative studies of polymeric nacre and nacre biomaterials	223
7.2 Conclusions.....	225
7.3 Future scope.....	226

List of Publication and Biography.....	227
---	------------

List of Figures

CHAPTER 1

Figure 1.1: Stereoisomer of lactic acid [2].....	1
Figure 1.2: Ring-opening polymerization of lactide dimer [2].....	1
Figure 1.3: Different foam morphologies and their applications.....	5
Figure 1.4: Crystalline porous microstructure of human bone [9].....	6
Figure 1.5: Nucleating effect of nanoclay on foam morphology: a) Schematic of polymer-clay network and the bubble nucleation effect, b) Foam morphology of PLA/clay nanocomposites whereas the inset shows the foam morphology of neat PLA [29].	12
Figure 1.6: The percentage of increase in the elastic modulus of polymer nanocomposite foams due to the addition of various Nanofillers [30].	13
Figure 1.7: Effect of D-lactic unit concentration on the spherulitic growth rate of poly(LL-co-meso-lactide)(open symbols, $M_n \approx 65$ Kg/mol) and poly(LL-co-DD-lactide) (filled symbols, $M_n \approx 74$ Kg/mol) [32].	16
Figure 1.8: Isothermal crystallization kinetics of PLA, nucleated PLA, nucleated and plasticized PLA [46].	18
Figure 1.9: Lamellar crystalline porous microstructure of rigid biostructures: a) Enamel of human teeth [59,60], b) Nacre's microstructure [61], c) Human bone's microstructure [62].	21
Figure 1.10: Crystalline microstructure of nacre: a) Protein templated/directed layer-by-layer 2-D crystallization mechanism of nacre [67], b) 2-D crystallized aragonite tablets [68], c) "Brick-and-mortar" crystalline microstructure [69], d) Nanoporous amorphous calcium carbonate (ACC) interlayers [69].....	22
Figure 1.11: Classical nucleation theory illustrating the critical radius size (r^*) for 3-D to 2-D nucleation mechanisms at corresponding S_2 & S_3 conditions. ($S_2 \gg S_3$).	27
Figure 1.12: Pressure-Temperature phase diagram of CO ₂ [88].....	29
Figure 1.13: Role of sc-CO ₂ in the polymer processing.	30
Figure 1.14: Solubility of CO ₂ in PLA: a) At low temperature region [97], b) Effect of d-content [98], c) At high temperature region [99].	32
Figure 1.15: Effect of CO ₂ plasticization on transition temperatures: a) Crystallization temperature, b) Glass transition temperature, c) Melting temperature [100].	33
Figure 1.16: a) Effect of CO ₂ pressure on the solubility of CO ₂ in PLA (PLA samples saturated at 30 °C with a contact time of 24 h) [101].	35
Figure 1.17: The calculated crystallization half-time of CO ₂ plasticized PLAs at different gas pressure conditions [100].	37

Figure 1.18: Schematic explaining various stages of foaming process [105].....	39
Figure 1.19: Plot of ΔG_{hom} Vs R	40
Figure 1.20: CO ₂ induced crystalline foam morphologies reported in the literatures: a) Layered nanoporous morphology [107], b) Layered porous morphology found in HDPE [108], c) Bubble nucleation around the PLA spherulites [109].	42
Figure 1.21: CO ₂ -induced conventional and non-classical crystallization process at saturation and supersaturation state respectively.....	44
Figure 1.23: Schematic of the plan of work. 49	

CHAPTER 2

Figure 2.1: Structure of PLA.	62
Figure 2.2: DSC scan of PLA-Neat.	63
Figure 2.3: TGA scan of PLA-Neat.....	64
Figure 2.4: FTIR spectra of PLA-Neat.	65
Figure 2.5: Structure of Cloisite 30B.....	66
Figure 2.6: WAXD analysis of Cloisite 30B.	67
Figure 2.7: TGA analysis of Cloisite 30B.	67
Figure 2.8: SEM micrograph of Cloisite 30B.....	68
Figure 2.9: Typical temperature run used for isothermal crystallization experiments.	71
Figure 2.10: Schematic of high pressure DSC cell enclosed with autoclave.....	76
Figure 2.11: A typical temperature run used for isothermal crystallization of CO ₂ plasticized PLA & PLANCs.	76
Figure 2.12: a) Experimental setup for CO ₂ induced crystallization process, b) Two-step CO ₂ crystallization process for inducing various degree of supersaturation conditions.	78
Figure 2.13: CO ₂ -induced crystallization process of PLA & PLANCs: a) High pressure vessel with temperature controller; b) Step change gas saturation profile used for CO ₂ -induced crystallization process.	79
Figure 2.14: Scheme of CO ₂ induced crystallization and foaming process of PLA and PLANCs.....	79
Figure 2.15: Sample preparation of CO ₂ induced crystallized PLANCCs: a) Laterally cryo-fractured surface, b) Vertically cryo-fractured surface.....	81
Figure 2.16: The key steps of image analysis of SEM micrographs	81

CHAPTER 3

Figure 3.1: TEM micrograph of PLANCs: a) PLANC-0.5, b) PLANC-1, c) PLANC-2.	86
Figure 3.2: Shear and diffusion induced delamination of clay platelets: (a) nanoclay stacks before melt intercalation, (b) nanoclay dispersion through polymeric chains diffusion, (c) nanoclay dispersion through shear intensity, (d) in reality, dispersion occurs through both mechanisms, although one is usually dominant.....	87
Figure 3.3: WAXD pattern of PLANCs with the inset showing waxd pattern of Cloisite 30B.	88
Figure 3.4: Complex viscosity vs frequency plot of PLA-Neat and PLANCs.	89
Figure 3.5: Storage modulus vs frequency plot of PLA-Neat and PLANCs with the inset showing the schematic of formation of polymer-clay network..	90
Figure 3.6: Cole-Cole plot of PLA-Neat and PLANCs.	91
Figure 3.7: Heat flow curves of PLA-Neat and PLANCs obtained from second heating scan of DSC.....	93
Figure 3.8: Cold crystallization curves of PLA-Neat and PLANCs.....	93
Figure 3.9: a) TGA curves of PLA-Neat and PLANCs, b) DTA curves of PLA-Neat and PLANCs.....	95
Figure 3.10: DMA analysis of thermally crystallized PLA-NC and PLANCCs: a) storage modulus vs temperature, b) $\tan \delta$ with function of temperature.	98
Figure 3.11: log-log plots of viscosity vs. Shear rate of PLA-Neat and PLANCs	99
Figure 3.12: Tensile stress versus strain curves of PLA-Neat and PLANCs.....	101
Figure 3.13: Impact properties of PLA-Neat and PLANCs.....	103

CHAPTER 4

Figure 4.1: Heat flow curves of PLA and PLANCs obtained at different isothermal temperatures a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.	107
Figure 4.2: Degree of crystallinity of PLA and PLANCs at different isothermal temperatures.	108
Figure 4.3: Relative crystallinity of PLA- and PLANCs at different isothermal temperatures a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.....	110
Figure 4.4: Avrami plots of PLA and PLANCs at different isothermal temperatures a) PLA-N, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.....	111

Figure 4.5: Overall crystallization rate (G) of PLA and PLANCs at different isothermal temperatures.....	113
Figure 4.6: Calculated Avrami index (n) of PLA and PLANCs at different isothermal temperatures.....	113
Figure 4.7: POM micrograph of isothermally crystallized PLANC films with the inset showing higher magnification micrograph (50X): a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.....	115
Figure 4.8: Effect of nanoclay on spherulitic density of PLANCs.....	116
Figure 4.9: Heat flow curves of PLA crystallized at different CO ₂ saturation conditions: a) 1 bar, b) 20 bar, c) 30 bar, d) 50 bar.....	117
Figure 4.10: Relative crystallinity of PLA obtained at different CO ₂ saturation conditions: a) 1 bar, b) 20 bar, c) 30 bar, d) 50 bar.....	119
Figure 4.11: Avrami plots of PLA at different CO ₂ pressure conditions: a) 1 bar, b) 20 bar, c) 30 bar, d) 50 bar.....	119
Figure 4.12: a) Effect of CO ₂ pressure on half crystallization time of neat PLA at different crystallization temperatures and b) Effect of CO ₂ pressure on overall crystallization rate (G) of PLA at a critical crystallization temperature of 100 °C.....	121
Figure 4.13: High pressure crystallization studies of PLA and PLANCs at various CO ₂ pressure conditions, P1= 1 bar, P2 = 20 bar, P3 = 30 bar, and P4 = 50 bar.....	123
Figure 4.14: Half-crystallization time of PLA & PLANCs crystallized at various CO ₂ pressure conditions.....	125
Figure 4.15: Overall crystallization rate, G of CO ₂ plasticized PLANCs at different pressure conditions studied at a critical crystallization temperature of 100 °C.....	126
Figure 4.16: Spherulite morphologies of PLANCs at various CO ₂ pressure conditions, P1= 1 bar, P2 = 20 bar, P3 = 30 bar, and P4 = 50 bar.....	127
Figure 4.17: Supersaturation driven crystal growth mechanism of PLA/CO ₂ system: a) Exclusion of CO ₂ from the crystal growth front creating supersaturation state at the crystal-amorphous interface, b) PLA/CO ₂ system is moving into supersaturated state by exclusion effect of CO ₂ and there is a formation of dendritic spherulitic structure as shown in the inset of figure.....	128

CHAPTER 5

Figure 5.1: Effect of gas saturation temperature on CO ₂ -induced crystallization of PLA. ..	134
Figure 5.2: Effect of nanoclay content on the gas solubility of PLANCs at 20 °C	136
Figure 5.3: (a) Experimental setup for CO ₂ induced crystallization process, and (b) Two-step CO ₂ crystallization process for inducing various degree of supersaturation conditions.	137

Figure 5.4: Classical nucleation theory shows the transition in nucleation mechanism from 3-D to 2-D and its effect on spherulitic microstructures.	138
Figure 5.5: In-situ observation of spontaneous nucleation and growth of PLA's spherulites at superheated temperature of 60 °C: a) 1s, b) 3s, c) 5s.....	139
Figure 5.6: Optical micrograph of CO ₂ induced spherulitic structures obtained at different supersaturation conditions: a) 60 °C, b) 70 °C, c) 75 °C, d) 80 °C.	140
Figure 5.7: Effect of superheated temperature on spherulites density.....	141
Figure 5.8: Comparison of spherulites density's values obtained at unsaturation, saturation, and supersaturation conditions.....	142
Figure 5.9: WAXD patterns of PLA films crystallized at different temperatures (60 °C, 70 °C, 75 °C, 80 °C).....	143
Figure 5. 10: SEM micrograph of cross-section plane of CO ₂ induced crystallized films: (a) 60 °C, (b) 65 °C, (c) 70 °C, and (d) 80 °C.....	144
Figure 5.11: POM micrographs of CO ₂ -induced crystallized PLA and PLANC's samples at superheated temperature of 80 °C: a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.	145
Figure 5.12: WAXD analysis of CO ₂ induced crystallized PLA & PLANCs at different superheated temperature of 80 °C.....	146
Figure 5.13: Figure 1: SEM images of cryofractured surfaces of cross-section planes of PLANCs: a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.....	147
Figure 5.14: CO ₂ -induced crystallization process: a) High pressure vessel with temperature controller; b) Step change gas saturation profile used for CO ₂ -induced crystallization process.....	150
Figure 5.15: Scheme of CO ₂ induced 2-D crystallization process and the investigation of 2-D spherulites morphology: a) Sample preparation, b) Crystalline morphology analyses.	151
Figure 5.16: SEM micrographs of self-stacked 2-D spherulitic microstructures: a) Inclined sectioned plane showing the self-stacking of disc-shape spherulites; b) Lateral sectioned plane showing positional disorder arrangement with inset showing the surface asperities...	152
Figure 5.17: Surface topology analysis: a) Schematic of 2-D spherulites and formation of cross-hatched lamellar pattern by the opposing lamellae, b) SEM micrographs of 2-D spherulitic morphologies of three successive planes	153
Figure 5. 18: a) Volume view of stacked layers with three layers; b) Result of orientation analysis.....	154
Figure 5.19: Optical profilometry images of two successive planes of layered structure; a) 2-D and b) 3-D images, c) Thickness profile of a typical sample in Z-direction.....	156
Figure 5.20: WAXD analysis: a) WAXD pattern of annealed PLA and CO ₂ -induced crystallized PLA, b) Typical peak fitting analysis of CO ₂ induced crystallized samples.....	157

Figure 5.21: Micro-sliced films of varying thickness: a) Light transmittance of self-organized 2D spherulites, b) Self-stacking of two dimensionally grown spherulites, c) Overlapping region from positionally disordered 2D spherulites, d) Cross-over region of two-adjacent spherulites.	160
Figure 5.22: Orientation analysis of layered lamellae.	161
Figure 5.23: SEM micrographs of cryo-fractured surfaces of CO ₂ -induced 2-D crystallized PLANC's samples prepared in the lateral direction: a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.	162
Figure 5.24: SEM micrographs of cryo-fractured surfaces of CO ₂ -induced 2-D crystallized PLANC's samples prepared in the transverse direction: a) PLA, b) PLANC-0.5, c) PLANC-1, d) PLANC-2.	163
Figure 5.25: WAXD analysis of CO ₂ induced 2-D crystallized PLA & PLANCs.	164
Figure 5.26: 2-D crystallization and its instantaneous self-templating assembly mechanism.	165
Figure 5.27: Man-made approach (Anisotropic assembly of building block) versus biomimetic crystallization approach: a) "Stack of coins" structure, b) "Stack of coins" crystalline microstructure made of 2-D spherulites.	165

CHAPTER 6

Figure 6.1: Lateral plane of CO ₂ -induced crystalline morphology of PLA samples obtained at different crystallization temperatures: (a) 120 °C, (b) 130 °C, (c) 140 °C, (d) 150 °C.	171
Figure 6.2: SEM analysis of stacked spherulite morphology: (a) Inclined section plane, (b) Radially sectioned plane, (c) Tangentially sectioned surface, (d) Typical open porous morphology observed from the amorphous sections between the 2-D spherulies.	172
Figure 6.3: Cell size distribution calculated from SEM micrograph using ImageJ.	173
Figure 6.4: Crystalline and foam morphological analysis of CO ₂ induced crystalline foamed sample (PLA-NCF): a) WAXD analysis of semi-crystalline 2-D spherulite, b) SAXS analysis of in-situ foamed inter-spherulitic layers, c) Snapshots of dye absorption test showing the absorption of a coloured hexane layer with time (dyed with Oil Red O by CO ₂ induced crystallized PLA).	174
Figure 6.5: (a) Schematic of self-stacked 2-D spherulites of PLA, (b) Schematic of brick-and-mortar microstructure, (c) SEM micrographs of developed brick-and-mortar crystalline microstructure of PLA.	175
Figure 6.6: SEM images of CO ₂ -induced self-organized crystalline layered microstructures obtained at four different crystallization temperatures: (a) 120 °C (b) 130 °C, (c) 140 °C, and (d) 150 °C with low and high magnification.	177

Figure 6.7: Effect of degassing temperature on 2-D crystalline microstructures: (a) Surface profile of cross-section surface, (b) Effect of degassing temperature on number of 2D spherulites and the appropriate change in corresponding thickness.	178
Figure 6.8: (a) WAXD patterns of PLA-NCFs obtained at different temperatures with the inset showing amorphous moulded samples, (b) Effect of degassing temperature on 2-D spherulitic microstructure and degree of crystallinity of CO ₂ -induced crystallized samples.	180
Figure 6.9: "Stack of coins" crystalline microstructure of PLANC-CFs: a) PLA-NCF, b) PLANC-0.5CF, c) PLANC-1CF, d) PLANC-2CF.	181
Figure 6.10: "Brick-and-mortar" crystalline microstructure of PLANC-CFs: a) PLA-NCF, b) PLANC-0.5CF, c) PLANC-1CF, d) PLANC-2CF with low and high magnification.	183
Figure 6.11: Crystallinity of CO ₂ induced crystallized PLANC-CFs measured from two different experiments: a) DSC (first scan), b) WAXD pattern.	184
Figure 6.12: a) TEM micrograph of 3-D crystallized PLANC-2C, b) TEM micrograph of 2-D crystallized PLANC-2CF.	185
Figure 6.13: Self association of nanoclay platelets and lamellar crystals: a) Convectional 3-D crystal growth process, b) non-classical 2-D crystal growth process.	187
Figure 6.14: Brick-and-mortar microstructures made of PLA biopolymer and nacre biomaterials with common design features: (a) Discontinuous laminate crystalline microstructure of PLA, (b) Discontinuous laminate crystalline microstructure of bivalve nacre,[13] (c) Nanoporous bridges in between 2-D PLA spherulites, (d) Nanoporous bridges in between the 2-D aragonite layers,[13] (e) Self-organized 2-D spherulitic structure, and (f) Self-organized 2-D crystalline platelets.[14]	188
Figure 6.15: (a) 2-D spherulites with exclusion of CO ₂ , (b) Self-templating assembly mechanism of 2-D spherulites with in-situ foaming of amorphous interlayers, (c) Perfect structural alignment of 2-D spherulites.	189
Figure 6.16: a) Layer-by-layer 2-D biomineralization approach of nacre, b) CO ₂ induced biomimetic crystallization approach with one-step assembly mechanism.	190
Figure 6.17: CO ₂ -induced 2-D crystallized impact test samples and their two different spherulites morphologies: a) Schematic of CO ₂ -induced 2-D crystallized impact test samples of PLA-NC, b) Schematic of CO ₂ -induced 2-D crystallized impact test samples of PLA-NCF, c) SEM micrograph of stack of 2-D spherulites with unfoamed inter-spherulitic amorphous region, d) SEM micrograph of stack of 2-D spherulites with nanoporous inter-spherulitic amorphous region.	194
Figure 6.18: Impact strengths and impact fracture mechanisms of CO ₂ -induced 2-D crystallized and foamed PLA samples: (a) Impact strength of PLA-NC and PLA-NCF, (b) Impact fracture surface of PLA-NC, (c) Impact fracture surface of PLA-NCF.	196
Figure 6.19: Impact fracture surfaces of polymeric nacreous crystalline microstructure of PLA: (a) SEM micrograph of Impact fractured surface of polymeric nacreous microstructure, (b) SEM micrograph of impact fracture surface at higher magnification, (c) SEM micrograph	

of cryofractured surface of lateral plane of impact tested samples, (d) Schematic representing bioinspired reinforcing mechanism found in nacre biomaterials with the inset showing shear displacement of disc shaped spherulites. 197

Figure 6.20: Flexural properties of CO₂-induced crystalline foamed samples (PLA-NCFs) with increasing density of 2-D spherulites..... 199

Figure 6.21: a) Flexural stress vs. deflection curve of thermal and CO₂ induced crystallize foamed samples showing flexural properties, b) Photograph of flexural PLA-NCF sample under bending stress, c) Schematic showing the deformation of 2-D spherulites at three different regions, d) SEM micrographs of deformed 2-D spherulites at corresponding three regions, e) Schematic of Instantaneous stress transfer and stress dissipation mechanisms... 200

Figure 6.22: SEM micrographs of deformed 2-D spherulites in the stair-case microstructure: a) SEM micrograph showing disc sliding mechanisms of 2-D spherulites in response to in-plane stress, b) SEM micrograph showing angular deformation of 2-D spherulites in response to in-plane stresses. 202

Figure 6.23: Schematic presenting two possible energy dissipation mechanisms of 2-D spherulites to in-plane stresses: a) Linear sliding of 2-D spherulites in response to in-plane stress, b) Angular sliding of 2-D spherulites in response to in-plane stress. 204

Figure 6.24: Percentage of increase in flexural strength of self-reinforced PLA vs PLA reinforced with natural fibres.[20]–[34] 204

Figure 6.25: DMA analysis of PLA-NCF and conventionally crystallized PLA-NC samples: a) Dynamic stiffness vs. temperature, b) tanδ vs. temperature. 205

Figure 6.26: Dynamic stiffness properties of crystallized PLANC-Cs & PLANC-CFs..... 207

Figure 6.27: DMA analysis of crystallized PLANC samples: a) Conventionally crystallized samples, b) CO₂ induced crystallized samples. 209

Figure 6.28: Flexural stress vs strain curves of CO₂ induced 2-D crystallized PLANC-CFs. 210

CHAPTER 7

Figure 7.1: Flowchart of investigations. 218

Figure 7.2: Self-organizing of 2D spherulites with perfect alignment and integration. 222

Figure 7.3: Biomineralization of nacre biomaterials versus CO₂-induced rapid biomimetic crystallization of polymeric nacre..... 224

List of Tables

CHAPTER 1

Table 1.1: Physical properties of PLA in comparison with other biopolymers [3].	2
Table 1.2: Comparison of material properties between PLA and PLA/clay nanocomposites [16].	9
Table 1.3: Effect of nanoclay on the thermal behaviour of PLANCS [18].	10
Table 1.4: Mechanical and thermal properties of amorphous and semi-crystalline PLA [27].	15
Table 1.5: Critical temperature and pressure values of different physical blowing agents [83].	29
Table 1.6: Rapid development of crystallinity in CO ₂ induced foamed PLA sample at their solid-state [98].	44

CHAPTER 2

Table 2.1: Materials specification of PLA 8052D©	62
Table 2.2: Thermal characterization of PLA-Neat.	64
Table 2.3: Characteristic peak list of PLA-Neat.	65
Table 2.4: Material specification of Cloisite 30B*	66
Table 2. 5: Specification of the twin screw extruder.	69
Table 2.6: Temperature profile of Twin screw Extruder during the melt compounding of PLANCS.	69
Table 2.7: Formulation of PLANCS.	70
Table 2.8: Injection moulding parameters.	72
Table 2.9: Compression moulding parameters.	74

CHAPTER 3

Table 3.1: Crystallization parameters obtained from non-isothermal DSC thermograms of PLANCS.	92
---	----

Table 3.2: TGA analysis of PLANCs.	94
Table 3.3: Power law index values of PLANCs.	98
Table 3.4: Tensile properties of PLANCs.....	99

CHAPTER 4

Table 4.1: Avrami crystallization parameters of PLA and PLANCs crystallization at different isothermal crystallization temperatures.	111
Table 4.2: Overall crystallization rate (G) of PLA-NCs at different isothermal temperatures.	116
Table 4.3: Avrami crystallization parameters of PLA crystallization at different CO ₂ saturation conditions.	120
Table 4.4: Isothermal crystallization kinetics of PLA and PLANCs at CO ₂ saturated conditions.....	122
Table 4.5: Avrami parameters of PLA and PLANCs crystallized at different CO ₂ pressure conditions.....	124

CHAPTER 5

Table 5. 1: Degree of crystallinity calculated from the CO ₂ -induced crystallized PLA.....	135
Table 5.2: Comparison of physical properties of thermally and CO ₂ -induced crystallized PLAs.....	158

CHAPTER 6

Table 6.1: Effect of nanoclay on 2D spherulitic stacking density.	182
Table 6.2: Flexural properties of CO ₂ induced 2-D crystalline foamed PLA-NCFs.	199
Table 6.3: Dynamic stiffness of 3-D and 2-D crystallized microstructures of PLANCs.	208
Table 6.4: Flexural properties of CO ₂ induced 2-D crystallized PLANC-CFs.....	211

List of Symbols

T_g	Glass transition temperature
T_c	Cold crystallization temperature
ΔH_c	Enthalpy of cold crystallization
T_m	Melting temperature
ΔH_m	Enthalpy of melting
T_m	Melt crystallization temperature
ΔH_m	Enthalpy of melt crystallization
ΔH_{mo}	Melting enthalpy of 100% crystalline PLA
X_c	Crystallinity from DSC
wt. %	Weight fraction
G'	Storage modulus
η^*	Complex viscosity
$\text{Tan } \delta$	Loss factor
K	Crystallization kinetic constant
$t_{1/2}$	Crystallization half-time (min)
G	Crystallization rate (1/min)
n	Avrami exponent
r^*_{2D}	Critical radius of disc shaped nucleus
r^*_{3D}	Critical radius of spherical nucleus
ΔG	Gibb free energy

List of Abbreviations

PLA	Polylactide
CO₂	Carbon dioxide
PLANCs	PLA/clay nanocomposites
PLANCCs	Crystallized PLANCs
PLANCCFs	Crystalline foamed PLANCs
SEM	Scanning electron microscopy
WAXD	Wide-angle X-ray Diffraction
SAXS	Small angle X-ray scattering
TEM	Transmission electron microscopy
POM	Polarized optical microscopy
DSC	Differential Scanning Calorimeter
HP-DSC	High pressure DSC
X_t	Relative crystallinity