

CHARACTERISATION OF HYDRATION OF SUPPLEMENTARY
CEMENTITIOUS MATERIALS WITH CALCIUM HYDROXIDE
FOR USE IN BINARY AND TERNARY CEMENTS

ANUJ



DEPARTMENT OF CIVIL ENGINEERING
INDIAN INSTITUTE OF TECHNOLOGY DELHI
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FOR USE IN BINARY AND TERNARY CEMENTS

by

ANUJ

DEPARTMENT OF CIVIL ENGINEERING

Submitted

in fulfilment of the requirements of the degree of DOCTOR OF PHILOSOPHY

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Dedicated to my Mother

CERTIFICATE

This is to certify that the thesis entitled “**CHARACTERISATION OF HYDRATION OF SUPPLEMENTARY CEMENTITIOUS MATERIALS WITH CALCIUM HYDROXIDE FOR USE IN BINARY AND TERNARY CEMENTS**”, being submitted by Mr. Anuj, to the Indian Institute of Technology Delhi, for the award of ‘Doctor of Philosophy’ in Department of Civil Engineering is a record of the bonafide research work carried out by him under my supervision and guidance. He has fulfilled the requirements for submission of this thesis, which to the best of our knowledge has reached the requisite standard.

The material contained in the thesis has not been submitted in part or full to any other University or Institute for the award of any other degree or diploma.

(Dr. Shashank Bishnoi)
Associate Professor
Department of Civil Engineering
Indian Institute of Technology Delhi
New Delhi – 110016, India
Date:
New Delhi

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ABSTRACT

The utilisation of supplementary cementitious materials (SCMs) and blended SCMs is widely promoted for use in cement and concrete. Although it is important to characterise the reactivity of SCMs before adding them to cement and concrete, the methods for measuring their reactivity have not been focused upon. Most of the reactivity test methods assess the reactivity in the presence of calcium hydroxide (CH), but the hydration behaviour of SCMs with CH and the effect of the factors that influence the hydration are not well understood. A better understanding of the test methods that measure reactivity of SCMs and their blends will help in improving their reliability and usability.

This thesis focuses upon understanding the hydration behaviour of SCMs and their blends for characterising their reactivity. While several test methods that quantify the reactivity of individual SCMs are available, the advent of ternary cements necessitates the testing of not just the individual SCMs, but also their blends. The hydration and reactivity of SCMs and their blends was studied in blends with CH and the optimum conditions, in terms of the availability and quantities of CH, sulphates, alkalis and their combinations, for the hydration of one each of calcined clay, fly ash and slag were studied. The kinetics of hydration, phase formation, CH consumption and strength development of selected SCMs and their blends were studied using isothermal calorimetry, X-ray diffractometry, thermogravimetry and mortar compressive strengths. The hydration results indicate that the presence of silica, alumina, calcium and magnesium influences the rate of hydration, and formation of hydration products. SCMs like calcined clay and slag react faster due to the presence of alumina and their hydration behaviour varies depending upon the presence of sulphates, alkalis and the temperature. Metastable hydroxy-AFm type phases were found to be unstable at high temperature and a reduction in compressive strength and bound water was observed due to their conversion to hydrogarnet (C_3AH_6).

Further, the reactivity of SCMs and their blends were measured with the help of 4 different methods: modified Chapelle's test for calcium consumption, chemical shrinkage test based on bound water, heat of hydration in the presence of CH, sulphates and alkalis and lime reactivity test based on strength development. The reactivity of SCMs and blends using the cement mortar compressive strength was tested for comparing the results from reactivity test methods with 28 days and 90 days compressive strength. The hydration and development of

hydrates in selected SCMs and blends were analysed under all the reactivity tests so as to understand the influence of the test conditions on the hydration reactions.

The results show that although various problems are associated with carrying out the reactivity tests both at ambient and increased temperatures, useful information can be obtained from these tests. While at ambient conditions, the rate of reaction of most SCMs is too slow to follow and characterise using standard techniques, at higher temperatures, the nature of the reactions and the products is seen to be influenced. In some cases, even phases that are usually inert, e.g. quartz, are seen to react when the temperature is increased. In some of the systems, phases such as hydrogarnet, that are usually not observed in cementitious systems at normal temperatures, were also found to be formed in the test methods. The test methods were still seen to provide important information like that capacity to react to portlandite, the relative reactivity and information about the kinetics of reactions.

This thesis presents the test methods used, the materials studied, the results obtained and the key conclusions derived from these test methods.

सारांश

सीमेंट और कंक्रीट में उपयोग के लिए पूरक सीमेंट सामग्री (सप्लमेंटरी सिमेंटीशियस मटीरीयल्ज - एस.सी.एम.) और मिश्रित एस.सी.एम. का उपयोग व्यापक रूप से किया जाता है। हालांकि एस.सी.एम. को सीमेंट और कंक्रीट में उपयोग करने से पहले उनकी अभिक्रियाशीलता को चिह्नित करना महत्वपूर्ण है, लेकिन उनकी अभिक्रियाशीलता को मापने के तरीकों पर ध्यान केंद्रित नहीं किया गया है। अधिकांश अभिक्रियाशीलता परीक्षण विधियों में कैल्शियम हाइड्रॉक्साइड (सी.एच.) की उपस्थिति में अभिक्रियाशीलता का आकलन किया जाता है, लेकिन सी.एच. के साथ एस.सी.एम. का जलयोजन व्यवहार और जलयोजन को प्रभावित करने वाले कारकों के प्रभाव को अच्छी तरह से नहीं समझा गया है। एससीएम और उनके मिश्रणों की अभिक्रियाशीलता को मापने वाले परीक्षण तरीकों की बेहतर समझ उनकी विश्वसनीयता और प्रयोज्य में सुधार करने में मदद करेगी।

यह थीसिस एस.सी.एम. और मिश्रित एस.सी.एम. के जलयोजन व्यवहार को समझने और उनकी अभिक्रियाशीलता को चित्रित करने पर केंद्रित है। जबकि कई परीक्षण विधियां जो विशिष्ट एस.सी.एम. की अभिक्रियाशीलता को निर्धारित करती हैं, उपलब्ध हैं, त्रिमिश्रित सीमेंट्स के आगमन के लिए न केवल अमिश्रित एस.सी.एम. के परीक्षण की आवश्यकता होती है, बल्कि मिश्रित एस.सी.एम. के परीक्षण की भी आवश्यकता होती है। एस.सी.एम. और मिश्रित एस.सी.एम. के जलयोजन और अभिक्रियाशीलता का अध्ययन सी.एच. के साथ और इष्टतम अवस्था वाले मिश्रणों में, जैसेकि सी.एच., सल्फेट्स, क्षार और उनके संयोजनों की उपलब्धता और मात्रा के संदर्भ में किया गया था। प्रत्येक निस्तापित क्ले, फ्लाइं ऐश और स्लैग में से एक-एक चयनित एस.सी.एम. का अध्ययन जलयोजन अभिक्रिया के लिए किया गया। चयनित एस.सी.एम. और मिश्रित एससीएम के जलयोजन गतिविज्ञान, अभिक्रिया उत्पाद, सी.एच. खपत और कम्प्रेसिव स्ट्रेंथ का अध्ययन समतापी कैलोरीमेट्री, एक्स रे डिफ्रेक्टोमेट्री, थर्मोग्रेवीमेट्री और मोटार कम्प्रेसिव स्ट्रेंथ का उपयोग करके किया गया था। जलयोजन परिणाम संकेत देते हैं कि सिलिका, एल्यूमिना, कैल्शियम और मैग्नीशियम की उपस्थिति जलयोजन की दर और अभिक्रिया उत्पादों के निर्माण को प्रभावित करती है। एल्यूमिना की उपस्थिति के कारण निस्तापित क्ले और स्लैग जैसे एस.सी.एम. तेजी से अभिक्रित होते हैं और सल्फेट, क्षार और तापमान की उपस्थिति के आधार पर उनका जलयोजन व्यवहार भिन्न होता है। मेटास्टेबल हाइड्रॉक्सी-एएफएम प्रकार के

उत्पादों को उच्च तापमान पर अस्थिर पाया गया और उनके हाइड्रैगनेट (C_3AH_6) में रूपांतरण के कारण कम्प्रेसिव स्ट्रेंथ और रासायनिक आबंध जल में घटौति देखी गई।

इसके अलावा, एस.सी.एम. और मिश्रित एस.सी.एम. की अभिक्रियाशीलता को ४ अलग-अलग तरीकों की मदद से मापा गया: कैल्शियम की खपत के लिए संशोधित चैपल परीक्षण, रासायनिक आबंध जल के आधार पर रासायनिक संकोचन परीक्षण, सी.एच. की उपस्थिति में जलयोजन की ताप, तथा सल्फेट्स, क्षार और सी.एच. की अभिक्रियाशीलता परीक्षण पर आधारित कम्प्रेसिव स्ट्रेंथ। एस.सी.एम. और मिश्रित एस.सी.एम. का उपयोग २८ दिनों और ९० दिनों की सीमेंट मोटार कंप्रेसिव स्ट्रेंथ के साथ अभिक्रियाशीलता परीक्षण के परिणामों से तुलना के लिए किया गया था। चयनित एस.सी.एम. और मिश्रित एस.सी.एम. में जलयोजन और जलयोजन उत्पादों के विकास का विश्लेषण सभी अभिक्रियात्मक परीक्षणों के तहत किया गया ताकि जलयोजन अभिक्रिया पर परीक्षण की स्थितियों के प्रभाव को समझा जा सके।

परिणाम बताते हैं कि हालांकि विभिन्न समस्याएं सामान्य और बड़े हुए तापमान पर अभिक्रियाशीलता परीक्षण करने से जुड़ी हैं, इन परीक्षणों से उपयोगी जानकारी प्राप्त की जा सकती है। जबकि व्यापक परिस्थितियों में, अधिकांश एस.सी.एम. की अभिक्रिया की दर मानक तकनीकों का उपयोग करने और उनका अनुसरण करने के लिए बहुत धीमी है, उच्च तापमान पर, अभिक्रियाशीलता और उत्पादों की प्रकृति को प्रभावित किया जाता है। कुछ मामलों में, यहां तक कि रासायनिक यौगिक जो आमतौर पर निष्क्रिय होते हैं, जैसे क्वार्ट्ज, तापमान बढ़ने पर अभिक्रियाशील देखा गया। कुछ परिस्थितिओ में में, हाइड्रोगनेट जैसे उत्पाद जो आमतौर पर सामान्य तापमान पर सीमेंटेड सिस्टम में नहीं देखे जाते हैं, अभिक्रियाशीलता परीक्षण विधियों में भी पाए गए थे। परीक्षण विधियों को पोर्टलैंडाइट से अभिक्रिया करने की क्षमता, सापेक्ष अभिक्रिया और अभिक्रियाशीलता के गतिविज्ञान जैसी महत्वपूर्ण जानकारी प्रदान करने के लिए देखा गया था।

यह थीसिस उपयोग किए गए परीक्षण विधियों, अध्ययन में इस्तेमाल की गई सामग्री, प्राप्त परिणामों और परीक्षण विधियों से प्राप्त मुख्य निष्कर्ष प्रस्तुत करती है।

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LIST OF ACRONYMS

AFm	Monosulphate
AFt	Ettringite
ASTM	American Standards for Testing and Materials
C-S-H	Calcium Silicate Hydrate
CC	Calcium Carbonate
CH	Calcium Hydroxide
C \bar{S}	Calcium Sulphates or Gypsum
DTG	Differential Thermogravimetry
EDS	Energy Dispersive Spectroscopy
EN	European Norm
FA	Fly Ash
Hc	Hemicarboaluminate
Htal	Hydrotalcite
IS	Indian Standard
LOI	Loss on Ignition
LS	Limestone
MAC	Mass Absorption Coefficient
Mc	Monocarboaluminate
MK	Metakaolin
NaOH	Sodium Hydroxide
OPC	Ordinary Portland Cement
PFA	Pulverised Fuel Ash
PPC	Portland Pozzolana Cement
PPC	Portland Pozzolana Cement
PSD	Particle Size Distribution
S	Slag
SCMs	Supplementary Cementitious Materials
SEM	Scanning Electron Microscopy
SF	Silica Fume
Si/Al	Ratio of Silica to Alumina
Si/Ca	Ratio of Silica to Calcium

SSA	Specific Surface Area
Str	Strätlingite
TGA	Thermogravimetry Analysis
w/c	Water to Cement Ratio
w/p	Water to Powder Ratio
XRD	X-Ray Diffraction
XRF	X-Ray Fluorescence

LIST OF ABBREVIATED FORMULAE

A	Al_2O_3
AFm	$\text{C}_6\text{A}\bar{\text{S}}_3\text{H}_{18}$
AFt	$\text{C}_6\text{A}\bar{\text{S}}_3\text{H}_{32}$
C	CaO
CC	CaCO_3
CH	$\text{Ca}(\text{OH})_2$
$\text{C}\bar{\text{S}}$	CaSO_4
H	H_2O
h-AFm	C_4AH_{13}
Hc	$\text{C}_3\text{A}.0.5\text{CC}_2\text{H}_{12}$
Hgt	C_3AH_6
Htal	$\text{Mg}_6\text{Al}_2(\text{OH})_{16}\text{CO}_3\cdot 4\text{H}_2\text{O}$
Mc	$\text{C}_3\text{A}. \text{CC}. \text{H}_{12}$
S	SiO_2
Str	C_2ASH_8