

PHOSPHORUS DIFFUSION IN
GALLIUM ARSENIDE

by

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ABSTRACTPHOSPHORUS DIFFUSION IN GALLIUM ARSENIDE

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The diffusion of phosphorus in gallium arsenide has been studied with the purpose of understanding the kinetics and mechanism of the diffusion process. The gallium arsenide phosphide formed after diffusion may then be used for the fabrication of various devices, e.g., light emitting diodes, power rectifiers, etc.

The conversion of gallium arsenide into gallium arsenide phosphide after phosphorus diffusion, was confirmed by Debye-Scherrer X-ray analysis of the diffused powder specimen. The Laue back reflection pattern of the surfaces of the diffused specimens showed that the diffused layer retains the crystallinity of the parent specimens.

The junction depth measurements taken on the diffused specimens prepared under different conditions, revealed that the diffusion process is suppressed considerably, when the diffusion is carried out under an overpressure of arsenic. This indicated that the

diffusion process is aided by vacancies on arsenic sites in gallium arsenide. The energy of formation of a vacancy at arsenic or phosphorus site in gallium arsenide phosphide was then estimated to be (1.06 ± 0.54) eV from these measurements. The behaviour of arsenic vacancies in gallium arsenide was also studied. It was found that these vacancies are electrically active.

The distribution of phosphorus atoms was determined by taking reflectance measurements at different depths of the diffused region. The activation energy and the pre-exponential factor of phosphorus diffusion in GaAs, as computed from the known distributions of phosphorus atoms in the specimens, were found to be (0.90 ± 0.04) eV and 7.0×10^{-9} cm²/sec. for GaAs and (1.51 ± 0.26) eV and 8.0×10^{-6} cm²/sec. for GaP, respectively. A diffusion mechanism **has been postulated** on the basis of the results obtained.

The changes occurring in electrical parameters, such as resistivity, mobility and carriers concentration; at different compositions of gallium arsenide phosphide (prepared by diffusion), were also measured. The ohmic contacts made on the surfaces of the specimens for electrical measurements showed dependence on the thickness of the alloy used for the purpose and also on the annealing temperature.

It was concluded that gallium arsenide phosphide prepared by diffusion at temperatures $\leq 800^{\circ}\text{C}$, may be suitable for the fabrication of various devices.

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