

**FUNCTIONAL HAIRY CORE-SHELL NANOFIBERS VIA  
BLOCK COPOLYMER SELF-ASSEMBLY**

**SUNITA SANWARIA**



**DEPARTMENT OF TEXTILE TECHNOLOGY  
INDIAN INSTITUTE OF TECHNOLOGY DELHI**

**AUGUST 2015**

© Indian Institute of Technology Delhi (IITD), New Delhi, 2015

**FUNCTIONAL HAIRY CORE-SHELL NANOFIBERS VIA  
BLOCK COPOLYMER SELF-ASSEMBLY**

by

**SUNITA SANWARIA**

**Department of Textile Technology**

Submitted  
in fulfilment of the requirements of the degree of  
**Doctor of philosophy**

to the



**Indian Institute of Technology Delhi**

**AUGUST 2015**

## Certificate

This is to certify that the thesis titled '*Functional Hairy Core-Shell Nanofibers via Block Copolymer Self-Assembly*' being submitted by **Ms. Sunita Sanwaria** to the Indian Institute of Technology Delhi, for the award of *Doctor of Philosophy* degree, is a record of bonafide research work carried out by her. She has worked under our guidance and supervision and fulfilled the requirements for the submission of thesis which has attained the standard required for a PhD. Degree of this Institute. The work carried out to complete this thesis has not been submitted for the award of degree or diploma in any institute in part or full.

**(Dr. Bhanu Nandan)**

Assistant Professor

Department of Textile Technology

Indian Institute of Technology Delhi

Date:

Place:

**(Dr. Rajiv Srivastava)**

Assistant Professor

Department of Textile Technology

Indian Institute of Technology Delhi

Date:

Place:

## ACKNOWLEDGEMENTS

Firstly I wish to express my gratitude to my supervisors Dr. Bhanu Nandan and Dr. Rajiv Srivastava, Department of Textile Technology, IIT Delhi, for supporting the idea and giving me opportunity to undertake this work. Their motivation skills, constant guidance, timely suggestions always helped me during the tenure of this research work. I shall be grateful to them for their invaluable suggestions and untiring help, which they have shown right from the beginning till the successful completion of this endeavor.

I am thankful to the Head, Department of Textile Technology for providing infrastructure facility and cooperation for smooth completion of this research work. I would also like to express thanks to my SRC members: Prof. Manjeet Jassel, Prof. Mangla Joshi, and Dr. Sameer Sapra for their valuable inputs and suggestions during the course of this work.

I am thankful to Prof. Manfred Stamm and Dr. Andriy Horechyy, Leibniz Institute for Polymer Research, Dresden, Germany for providing access to microscopy facilities. I am also thankful to Prof. Hsin-Lung Chen, Department of Chemical Engineering, National Tsing-Hua University, Taiwan for helping in the SAXS measurements.

I am thankful to Department of Chemistry, IIT Delhi for providing UV-Visible Spectroscopy facility. I am also thankful to AIRF Centre (Jawaharlal Nehru University, Delhi) and SAIF (All India Institute of Medical Sciences, Delhi) for providing TEM facility.

I am thankful to Mr. V.K Kala, Mr. Amarjeet, Mr. A.K. Sehgal and other members of the textile department for their support.

I would also like to acknowledge my college, Deshbandhu College, for allowing me to undertake the Ph.D. work and grant leave for higher studies. I am thankful to my fellow colleagues Mythili and Harika for encouraging me all the time. I extend my gratitude to my colleagues Mr. Jit Pal and Mr. Sajan Singh for their willingness to help in any situation. I am also thankful to Pratick, Archana, Esha, Thangapandian and all other members of our research group for their constant support and encouragement.

I owe thanks to my parents for their support, help and understanding during the entire period of this research work. And special thanks to my sister Ms. Swadesh Sanwaria for her immense support during the entire tenure.

Lastly, I express my sincere thanks to all those who have helped me directly or indirectly during the course of my research work.

**Sunita Sanwaria**

# Abstract

Recently, a lot of interest has been shown for the synthesis of functional shaped polymer nano-objects due to their potential application in tissue engineering, drug delivery, catalyst carrier, fillers for polymer matrices etc. One of the convenient approach to prepare such polymer nano-objects is *via* block copolymer self-assembly. In this case, the self-assembled domains of block copolymer could be isolated as individual nano-objects by selectively swelling the matrix phase of the block copolymer structure. In the present work, we demonstrate this concept using a cylinder forming polystyrene-*block*-poly(4-vinylpyridine) (PS-*b*-P4VP) block copolymer. The PS cylindrical domains were isolated as individual nanofibers by swelling the P4VP matrix in methanol which is a non-solvent for PS chains. The isolated nanofibers (cylinders) had a core-shell structure with PS as the rigid core and P4VP blocks constituting the hairy shell. The nanofiber diameter could be easily tailored by either varying the molecular weight of the block copolymer and/or by mixing PS homopolymer with the PS-*b*-P4VP block copolymer. Furthermore, the nanofibers could also be transformed to core-shell nanospheres either by aging or heating the nanofiber dispersion.

The core-shell nanofiber/nanosphere were further used as templates to direct the assembly of functionalities such as inorganic nanoparticles. In this case, the P4VP shell could easily be used as host because of the excellent reactivity of pyridine group towards inorganic surfaces. This was demonstrated by loading gold (Au) and cadmium sulphide (CdS) nanoparticles on P4VP shell of the nanofiber. Moreover, by directly performing the sol-gel chemistry on the P4VP shell and later on removing the organic phase, pure silica and titania nanotubes and hollow nanospheres were also fabricated. The inert PS core of the nanofibers were also used as host for the nanoparticles by suitably functionalizing the nanoparticles with oligomeric PS shell. The silver (Ag) nanoparticles incorporated this way in the PS core were found to pack in an interesting helical symmetry. It was further demonstrated that both core and shell of the nanofibers could be loaded with two different targeted functionalities which enables the combination of their properties in a single structure. This was aptly shown by fabricating Ag/Au and Ag/CdS multifunctional polymer nanofibers where the Ag nanoparticles were in the core and Au or CdS nanoparticles decorated the shell of the nanofibers.

# Contents

<b>Certificate</b>	i
<b>Acknowledgments</b>	ii
<b>Abstract</b>	iv
<b>List of figures</b>	ix
<b>List of symbols and abbreviations</b>	xiv
<b>Chapter 1 Introduction and Motivation</b>	1
<b>1.1 Introduction</b>	1
<b>1.2 Goal of the work</b>	4
<b>1.3 Thesis outline</b>	4
<b>Chapter 2 Theoretical background and Literature Overview</b>	7
<b>2.1 Block Copolymers</b>	7
<b>2.2 Polymer Nano-objects</b>	12
2.2.1 Synthetic Approach	13
2.2.2 Block copolymer Self Assembly Approach	17
<b>2.3 Polymer Nano-objects as Templates</b>	28
<b>Chapter 3 Experimental Techniques</b>	41
<b>3.1 Scanning Electron Microscopy</b>	41
<b>3.2 Transmission Electron Microscopy</b>	43
<b>3.3 Energy Dispersive X-ray Analysis</b>	46
<b>3.4 Atomic Force Microscopy</b>	47
<b>3.5 Small angle X-Ray Scattering</b>	49
<b>3.6 UV-visible spectroscopy</b>	51
<b>3.7 Dynamic Light Scattering</b>	52

<b>Chapter 4 Fabrication of Polymer Nano-objects</b>	<b>53</b>
<b>4.1 Introduction</b>	<b>53</b>
<b>4.2 Experimental Section</b>	<b>56</b>
4.2.1 Preparation of Polymer Nanofibres	56
4.2.2 Tuning the Diameter of Nanofibers	68
4.2.3 Fabrication of Nanospheres from Nanofibers	59
4.2.4 Characterization	59
<b>4.3 Result and Discussion</b>	<b>61</b>
4.3.1 Polymer Nanofibers	61
4.3.2 pH Responsive behavior of Nanofibers	65
4.3.3 Nanofibers with tuned diameter	66
4.3.4 Polymer Nanospheres from Nanofibers	71
<b>4.4 Conclusion</b>	<b>74</b>
<b>Chapter 5 Polymer Nano-objects Shell as Host for Functionalities</b>	<b>77</b>
<b>5.1 Introduction</b>	<b>77</b>
<b>5.2 Experimental Section</b>	
5.2.1 Fabrication of Metal oxide/polymer hybrid Nano-objects	79
5.2.2 Fabrication of Metal/polymer hybrid Nano-objects	82
5.2.3 Characterization	83
<b>5.3 Result and Discussion</b>	<b>84</b>
5.3.1 Fabrication of Metal oxide/polymer hybrid Nano-objects	84
5.3.2 Fabrication of Metal/polymer hybrid Nano-objects	92
<b>5.4 Conclusion</b>	<b>98</b>
<b>Chapter 6 Polymer Nano-objects Core as Host for Functionalities</b>	<b>100</b>
<b>6.1 Introduction</b>	<b>100</b>

<b>6.2. Experimental</b>	102
6.2.1 Synthesis of Polystyrene-coated silver nanoparticles	102
6.2.2 Synthesis of Nanofibers of PS-b-P4VP with silver in the core	103
6.2.3 Characterization	103
<b>6.3 Result and Discussion</b>	105
6.3.1 Analysis of synthesized nanoparticles	105
6.3.2 Block Copolymer/Nanoparticle Bulk morphology	106
<b>6.4 Conclusion</b>	119
<b>Chapter 7 Fabrication of Multifunctional Nanofibers</b>	121
<b>7.1 Introduction</b>	121
<b>7.2 Experimental Section</b>	122
7.2.1 Fabrication of Multifunctional Nanofibers	122
7.2.2 Characterization	123
<b>7.3 Result and Discussion</b>	124
7.3.1 Fabrication of Ag/CdS/polymer hybrid Nanofibers	124
7.3.2 Fabrication of BCP Ag/Au hybrid Nano-objects	127
<b>7.4 Conclusion</b>	131
<b>Chapter 8 Summary and Future Outlook</b>	133
<b>8.1 Summary</b>	133
<b>8.2 Future Outlook</b>	135

# List of Figures

**Figure 2.1** Different kinds of block copolymer.

**Figure 2.2** Illustration of change in curvature as volume fraction of diblock copolymer changes.

**Figure 2.3** Dependence of the observed morphologies on  $\chi N$  and  $f$ .

**Figure 2.4** The Phase Diagram of ABC triblock copolymer.

**Figure 2.5** Strategies for the synthesis of cylindrical polymer brushes.

**Figure 2.6** Synthetic procedure for amphiphilic core-shell CPBs with a hydrophilic PAA core and a hydrophobic PnBA shell.

**Figure 2.7** Various structures of cylindrical polymer brushes.

**Figure 2.8** Various self-assembled structures formed by amphiphilic block copolymers in a block-selective solvent.

**Figure 2.9** Transmission electron microscopy (TEM) micrographs and corresponding schematic diagrams of various morphologies formed from amphiphilic PSm-b-PAA<sub>n</sub> copolymers.

**Figure 2.10** Isolated nano-objects from bulk microphase separated structures.

**Figure 2.11** TEM images of PS-b-PI nanofiber fraction obtained after chemical crosslinking.

**Figure 2.12** Ordered Organic/Inorganic Hybrid Materials by Self-Assembly of Gelation Block Copolymer PTEPM-b-PS in Bulk and Isolated Hybrid Plate and Cylinder.

**Figure 2.13** Preparation of P2VP Functionalized Organic/Inorganic Hybrid Nano-objects with Different Morphologies and Organization of Gold Nanoparticles along Surfaces of the Hybrid.

**Figure 2.14** JPs of different architectures prepared via the selective crosslinking of the middle block of triblock terpolymer bulk phases and subsequent dissolution.

**Figure 2.15** Schematic of the metallization route, starting from PS nanorods covered with P4VP.

**Figure 2.16** Illustration showing the synthetic strategy toward the nanostructured particles through self-assembly of PS-b-P4VP(PDP)<sub>x</sub> under 3D soft confinement as well as the favorable growth of other functional species onto the nano-object surface.

**Figure 2.17** Synthesis of Template CPBs and Their Use in the Template-Directed Synthesis of 1D Silica Hybrid Nanostructures. TEM micrographs of the as-synthesized silica hybrid nanowires.

**Figure 2.18** Synthetic route to obtain water-soluble organo-silica hybrid nanotubes templated by core-shell-corona structured CPBs.

**Figure 2.19** Sequential anionic polymerization of polystyrene- block -poly(allyl methacrylate) (PS- b -PAMA) into hexagonally packed cylinders to produce anisotropic metal oxide–polymer brush hybrid nanowires. TEM images of as-synthesized titania nanorods obtained using nanofibers, isolated from PS-b-PAMA, as templates.

**Figure 2.20** Synthesis of titania-CPB hybrid nanowires and inorganic titania nanowires.

**Figure 2.21** Preparation of Well-Defined Rare-Earth Metal Cations ( $\text{Ln}^{3+}$ ) Incorporated Silica Hybrid Nanoparticles via the Template-Directed Approach Based On Core-Shell Cylindrical Polymer Brushes.

**Figure 2.22** TEM micrographs of the hybrid nano-objects loaded with gold nanoparticles dispersed in water (pH = 3).

**Figure 2.23** Illustration of comb-coil supramolecular assembly (SMA) based on PS-b-P4VP(PDP)<sub>1.0</sub> ( $\phi(\text{PS})=23\%$ ), the hierarchical structures formed by co-assembly of the PS-b-P4VP(PDP) SMA and PS-grafted Au NPs, and isolated hybrid cylindrical micelles obtained by solvent-assisted disassembling of the Au NP-loaded PS-b-P4VP(PDP) SMA.

**Figure 3.1** Schematic diagram of Scanning Electron Microscopy Instrument.

**Figure 3.2** Schematic diagram of Transmission Electron Microscope Instrument.

**Figure 3.3** Schematic diagram of Atomic Force Microscopy (a) Instrumentation (b) Principle involved in the Atomic Force Microscopy showing Force versus distance curve and AFM operating modes.

**Figure 4.1** Schematic Illustration of fabrication of Nanofibers.

**Figure 4.2** Schematic Illustration of fabrication of Nanospheres from Nanofibers.

**Figure 4.3** SAXS profile of bulk sample of PS-b-P4VP.

**Figure 4.4** Transmission Electron Micrograph of the bulk sample.

**Figure 4.5** TEM and SEM image of drop casted sample.

**Figure 4.6** AFM topography image of spin-coated polymer nanofibers prepared from PS-b-P4VP on the silicon wafer (a) Height image and (b) Phase image.

**Figure 4.7** Turbidity showing the pH response of PS-b-P4VP nanofiber in aqueous dispersion.

**Figure 4.8** UV spectrum of the nanofiber dispersion in aqueous medium at pH 3 and 8.

**Figure 4.9** SAXS profile of the nanofibers tuned by using the block copolymer of lower molecular weight.

**Figure 4.10** Micrographs of nanofibers images after tuning their diameter by using block copolymer of lower molecular weight PS<sub>10400</sub>-b-P4VP<sub>19200</sub>. (a) SEM image (b) TEM image.

**Figure 4.11** Schematic Illustration of different regimes in BCP/homopolymer mixture as the function of homopolymer molecular weight.

**Figure 4.12** SAXS profile of the nanofibers tuned by using the mixture of block copolymer and homopolymer.

**Figure 4.13** SEM images after tuning the diameter of the nanofibers (a, c). TEM images after tuning the diameter of the nanofibers (b, d)

**Figure 4.14** Nanospheres (a) SEM image (b) TEM image.

**Figure 4.15** Size distribution histogram of polymer nanosphere obtained from DLS measurements.

**Figure 5.1** Schematic for the ex-situ as well as in-situ approach for functionalizing the nanofibers.

**Figure 5.2** TEM image of polymer nanofiber after (a) silica deposition and (b) after calcination.

**Figure 5.3** EFTEM images of silica deposited polymer nanofibers showing silicon map, oxygen map, carbon map.

**Figure 5.4** EFTEM images of silica deposited polymer nanofibers after calcination showing the distribution of different elements in the nanofibers.

**Figure 5.5** TEM image of silica/polymer hybrid nanospheres before and after calcination.

**Figure 5.6** EFTEM images of silica deposited polymer nanospheres shows the distribution of different elements in the nanofibers.

**Figure 5.7** EFTEM images of silica deposited polymer nanospheres showing the distribution of different elements in the nanofibers.

**Figure 5.8** TEM micrographs a) hybrid / polymer titania nanofiber b) pure titania nanofiber.

**Figure 5.9** EFTEM images of silica deposited polymer nanofibers showing the distribution of different elements in the nanofibers: carbon map, titanium map, oxygen map.

**Figure 5.10** EFTEM micrographs after calcination of titania/polymer hybrid nanofiber showing the elemental mapping titanium, oxygen and carbon.

**Figure 5.11** SEM micrographs a) hybrid / polymer titania nanospheres b) pure titania nanospheres.

**Figure 5.12** EFTEM micrographs showing the mapping of elements (a-c) before and (d-f) after calcination of titania/polymer hybrid nanospheres.

**Figure 5.13** XRD spectrum of the hybrid nanofibers after calcination.

**Figure 5.14** SEM images of gold/polymer hybrid nanofibers by in-situ and ex-situ approach

**Figure 5.15** Polymer nanofibers coated with gold nanoparticles (a) TEM image (b) EDX spectra

**Figure 5.16** UV-visible spectra of the hybrid gold/polymer nanofibers.

**Figure 5.17** TEM image of polymer nanofibers coated with CdS nanoparticles.

**Figure 5.18** EDX spectra of gold-polymer hybrid nanorods.

**Figure 5.19** UV-Visible spectra of the hybrid CdS/polymer.

**Figure 5.20** Plot exhibiting the band gap of the CdS nanoparticles.

**Figure 6. 1** (a) TEM image and (b) size distribution histogram of AgNP.

**Figure 6.2** FTIR plot of PSSH and PSSH-stabilized Ag nanoparticles.

**Figure 6.3** SAXS profile of the neat block copolymer and its composite with 10wt% AgNP.

**Figure 6.4** SAXS scattering profile of PS-b-P4VP/AgNP@10wt% composite where the intraparticle scattering observed from the AgNP has been fitted using the sphere form factor.

**Figure 6.5** TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with (a) 4 wt.% of AgNP, (b) 8 wt.% of AgNP.

**Figure 6.6** TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with 10 wt.% of AgNP (a) view along the plane of the PS cylinder axis, (b) view normal to the plane of PS cylinder axis.

**Figure 6.7** TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with 20wt. % of AgNP.

**Figure 6.8** (a) TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with 10 wt. % of AgNP (b) TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with 8 wt.% of AgNP (c) TEM images of self-assembled morphologies formed in AgNP/PS-b-P4VP sample loaded with 10 wt.% of AgNP .

**Figure 6.9** Bright field TEM and EFTEM images of NF@Ag10% sample displaying nitrogen, carbon and silver elemental maps.

**Figure 6.10** UV spectra of Ag nanoparticle and BCP nanofibers with Ag nanoparticle in core.

**Figure 6.11** 3D reconstruction obtained by tomography of AgNP packing inside the NF formed by the PS-b-P4VP block copolymer (BCP).

**Figure 6.12** (a) TEM image of ultramicrotomed bulk sample of PS-b-P4VP/AuNP@4wt% with AuNP with core size of ~5 nm (TEM). The AuNPs are predominantly located at PS/P4VP interface; (b) TEM image of ultramicrotomed bulk sample of PS-b-P4VP/AgNP@4wt% prepared from BCP of lower molecular weight ( $M_n(\text{PS}) = 10400 \text{ g/mol}$ ,  $M_n(\text{P4VP}) = 19200 \text{ g/mol}$ ,  $\phi(\text{PS}) \sim 0.35$ ) and AgNP with core size of 6.9 nm

**Figure 7.1** Nanofibers with silver nanoparticle in the core and CdS nanoparticles on the shell.

**Figure 7.2** EDX spectra of multifunctional nanofiber having Ag in core and CdS in the shell showing carbon, silver, cadmium and sulphur.

**Figure 7.3** UV-visible spectral comparison of nanofibers with CdS on shell, nanofibers with Ag in core and nanofibers with Ag and CdS.

**Figure 7.4** Nanofibers with silver nanoparticle in the core and CdS nanoparticles on the shell.

**Figure 7.5** EDX mapping of different metal on the nanofibers viewed under TEM.

**Figure 7.6** UV-visible spectral comparison of nanofibers with Au on shell, nanofibers with Ag in core and nanofibers with Ag and Au.

## List of Symbols and Abbreviations

PS- <i>b</i> -PMMA	polystyrene-block-poly(methyl methacrylate)
N	degree of polymerization
X	Flory-Huggins interaction parameter
f	volume fraction
PS- <i>b</i> -PI	polystyrene-block-polyisoprene
CPB	cylindrical polymer brushes
BCP	block copolymer
ATRP	atom transfer radiated polymerization
ROP	ring opening polymerization
NMP	nitroxide mediated polymerization
MWD	molecular weight distribution
PBIEM	poly(2-(2-bromoisobutyryloxy)ethyl methacrylate)
PS	polystyrene
PAA	polyacrylic acid
P <i>t</i> BA	poly( <i>t</i> -butyric acid)
PCL	polycaprolactam
PDMAEMA	poly(N, N-dimethylaminoethyl methacrylate)
PS- <i>b</i> -PCEMA	polystyrene-block-poly(2-cinnamoyloxyethyl methacrylate)
PBMA	poly(butyl methacrylate)
PS- <i>b</i> -P4VP	polystyrene-block-poly(4-vinylpyridine)
PS- <i>b</i> -P2VP	polystyrene-block-poly(2-vinylpyridine)
PTEPM- <i>b</i> -PS	poly 3-(triethoxysilyl)propyl methacrylate block-polystyrene
P <i>t</i> BA- <i>b</i> -PGMA	poly( <i>tert</i> -butyl acrylate)-block-poly(glycidyl methacrylate)

EDA	ethylenediamine
PA	propargylamine
PI-b-PEO	polyisoprene-block-poly(ethylene oxide)
PDP	3-n-pentadecylphenol
POEGMA	poly(oligo (ethylene glycol) methacrylate)
TALH	titanium (IV) bis(ammonium lactate) dihydroxide
PHEMA	poly(2-hydroxyethyl methacrylate)
PMAEMA	poly(2 methylamino)ethylmethacrylate
$\text{In}^{+3}$	lanthanum ion
SEM	scanning electron microscope
TEM	Transmission electron microscope
EFTEM	Energy filtered Transmission electron microscope
DLS	Dynamic light Scattering
SAXS	Small angle x-ray scattering
AFM	Atomic force microscopy
TEOS	tetraethylorthosilicate
TTIP	titanium isopropoxide
CdS	cadmium sulphide
PS- b -PAMA	poly(allyl methacrylate)
PSS	polystyrene sulfonic acid
$\text{NaBH}_4$	Sodium borohydride
Au	gold
Ag	silver
SPR	surface plasmon resonance
hPS	homopolystyrene